TYBSC SEM VI PHYSICAL CHEMISTRY (MODEL MCQ's)

| | |
|---|---|
| 1 | Which of the following is a uni-univalent electrolyte?<br>a) KCl<br>b) $CaCl_2$<br>c) $AlCl_3$<br>d) $MgCl_2$ |
| 2 | The ionic strength of a solution of molality $m_i$ and valency of the ion Zi is ____<br>a) $\mu = \frac{1}{2} m_i Zi^2$<br>b) $\mu = \sum m_i Zi^2$<br>c) $\mu = \frac{1}{2} \sum Zi^2$<br>d) $\mu = \frac{1}{2} \sum m_i Zi^2$ |
| 3 | Which of the following is an example of electrolyte concentration cell with transference?<br><br>a) Cu \| $CuSO_4$ \|\| $AgNO_3$ \| Ag<br>b) Pt , $H_{2(g)}$ \| $HCl_{(aq)}$ \| $AgCl_{(s)}$ Ag<br>c) Pt , $H_{2(g)}$ \| $HCl_{(aq)}(a_1)$ \|\| $AgNO_3(a_2)$ \| Ag<br>d) Ag – AgCl \| NaCl $(a_1)$ \| NaCl$(a_2)$ \| AgCl - Ag |
| 4 | A concentration cell which has salt bridge in its representation is ___<br>a) electrolyte concentration cell with transference<br>b) Electrode concentration cell<br>c) Electrolyte concentration cell without transference<br>d) Gas concentration cell |
| 5 | The total emf of the electrolyte concentration cell with transference containing the liquid junction potential is given by .....<br>a) $E_j = E_{oxidation} + E_{reduction}$<br>b) $E_{cell} = E_{oxidation} + E_{reduction} + E_j$<br>c) $E_{cell} = E_{oxidation} - E_{reduction} - E_j$<br>d) $E_j = E_{oxidation} - E_{reduction}$ |
| 6 | The activity 'a' is related to the activity coefficient '' $\gamma$' by the equation___<br>a) $a = \frac{m}{\gamma}$<br>b) $\gamma = a - m$<br>c) $a = m + \gamma$<br>d) $a = m\gamma$ |

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Bahl Arun/ Bahl B.S. & Tuli G.D.

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