#### FINAL EXAMANITION IN PHYSICAL SCIENCE

			as, choose the BEST answ t corresponds to the letter	
1.	The chemical form	rula of caffeine is Callinh	LO: What is its molecular m	0.007
		B. 172 g/mol		D. 203 g/mol
2.	People usually use	hydrogen peroxide (H <sub>1</sub> C	nolar mass- 34 02g/m	of) to clean their wounds. I
				many moles of H <sub>2</sub> O <sub>2</sub> dad h
	A. 0.035 mol	B. 0.030 mol	C. 0.029 mol	D. 0.0025 mol
3.	Methane (CH, molar mass = 16.04 g/mol) is one of the greenhouses gases. If 32.08 g of CH, we			
	released in the atmosphere, how many molecules of CHs were added in the atmosphere?			
	A. 1.20 30102	B. 1.20 X10 <sup>23</sup>	C. 6.02 X102*	D. 6.02 X10 <sup>23</sup>
Fo	r numbers 4 and 5.	consider the problem bel	ow.	
4	Iron (III) conde reacts with carbon monocode to form iron and carbon diconde. Which is the balanced chemical equation that represents the reaction?			
	A. Fe <sub>2</sub> O <sub>3</sub> + 3CO		C 2 Fe <sub>2</sub> O <sub>3</sub> + 3 CO	-> 4 Fe +3 CO-
	B Fact to	Fe CO.	D FeO + 3 CO ->	
40	What mass of iron	(III) conde is needed to ;	wadane 5 6 kg of iron?	15 5 5 5 5 5
40	A 40kg	B 80 kg	C. 12 kg	D 16 kg
	Jr. wong	D. O'C ME	0.12.06	D TORE
F	or numbers 6-8, con	sider the problem below:		
	Ammonium carbo	nate decomposes to form	ammonia gas, carbon dio	code and water. The reacti
repre	esented by the equat	ion below		
		$(NH_3)_0CO_3 \rightarrow 2 NF$	H <sub>3</sub> + CO <sub>2</sub> + H <sub>2</sub> O	
6	How many moles of ammonia are formed for every mole of the reactant used?			
	A. 1 mole	B. 2 moles	C. 2.5 moles	D. 3 moles
7.	How many moles of the reactant is required to produce 22 grams of CO <sub>2</sub> ?			
	A. 0.50 mole	B. 1.0 mole	C. 1.5 mole	D. 2.0 moles
3	How many grams	of (NH <sub>2</sub> ) <sub>2</sub> CO <sub>2</sub> is required	to produce 1.5 moles of	water?
	A 49 g	B 96 g	C. 144 g	D. 192 g
For	numbers 9 and 10	consider the problem be	low.	<del>-</del>
			on excess of HC1, yielding	48 25 g of VCI
		CO <sub>2</sub> _+ 2 HCl → 2 KC		, B
	What is the theore		, ,	
200	A 46.3 g	B 480g	C 48.3 g	D 495@
			C. +5.3 E	D. 49.3 g
10	What is the percer	if yneld?	Company and the Company of the Compa	The Court State of the Court of
4.0	A. 92.5%	B. 93.5 %	C. 97.0%	D. 97.5 %
11.	Which of the following will cause a soap to reduce its cleaning activity?			
	A Alkyl hydrogen sulfate C presence of surfactants			
4.00	B. Hard water D. soft water			
12	Which of the following chemicals are linked to breast cancer in women?			
	A. Parabens	B PEG	C. Paraffin wax	D BHA
1.3	Which of the follo	wing disinfectants has a	fungicidal property and is	a potential carcinogen?

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**Ransford Tetteh** 

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its discovery in 1911 to the present not as a history book in the usual sense but through close ups of the leading characters and their role in that story the Nobel laureates who were still among us in the years 2001 2004 when the main round of interviews was carried out Since then two of them already passed away For each one of the 10 laureates the author tells their story by direct quotation from interviews in their own words Each chapter treats one laureate The author first gives a brief account of the laureates scientific background and main contribution Then each laureate tells his own story in his own words Recent Advances in Heat Pipes Wael I.A. Aly, 2019-09-18 Heat pipes This book is unique in its approach to science history are considered as an effective thermal solution particularly in high heat flux applications and in situations where there is a combination of nonuniform heat loading limited airflow over the heat generating components and space or weight constraints This book is intended to explore some of the recent advances in heat pipes and their applications in thermal systems The first chapter is an introductory chapter about the recent advances in heat pipes in general The second chapter is about thermosyphon heat pipe technology working principles advantages and disadvantages application ranges and using computational fluid dynamics in modeling thermosyphons The third chapter is about recent research into loop heat pipes LHPs The last chapter presents a novel liquid vapor separator incorporated gravitational LHP Annual Report Resources in Education, 1994-11 Mpumalanga (South Africa). Department of Education, 2014

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