Element	symbol	Atomic number	Electronic arrangement in different orbit					
			К	L	M	N	О	P
Hydrogen	Н	1	1					
Helium	He	2	2					
Lithium	Li	3	2	1	un aren 8			
Beryllium	Be	4	2	2				
Boron	В	5	2	3				
Carbon	C	6	2	4	WB	BSES	olutio	ns.
Nitrogen	N	7	2	5				
Oxygen	0	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8			77174	
Sodium	Na	11	2	8	1			
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5			
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	К	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

 $\bullet \quad \overline{\mathbf{v}} = \frac{1}{\lambda} = \frac{\mathbf{v}}{c}$

[v = wave number]

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author correlates the development of ideas with trends in the Cultural Revolution and against this background it is possible to understand why debates over general philosophy gave way to conflicts over specific sciences in the aftermath of the first Five Year Plan and why there was a genuine crisis in Soviet biology

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