

Physical Science EOCT Study Guide – Domain 3

Physical Energy, Force, and Motion

Standards: PS.3.1 and 3.2

1. Students will analyze transformations and flow of energy within a system.
 - a. Explain thermal energy (heat) and kinetic energy (movement) and how they are related.
 - b. Explain heat. Thermal energy (heat) moves from molecules at a higher temperature to lower temperature.
 - c. Explain specific heat. The amount of energy needed to raise the temperature of 1 kg of substance by 1°C.
 - d. What is the formula for determining that change in heat that energy can hold?

$$Q = mc\Delta T \quad \Delta T = T_2 - T_1$$
 - e. A sample of 0.5 kg of water at 20°C and specific heat of 4186 J/kg°C. Find the change in heat energy if the block is 2 mmol from 10°C to 20°C.

$$Q = mc\Delta T \quad 20 = 20 - 10 = 10 \quad 4186, \text{ and } 0.5$$
 - f. Explain conduction, convection, and radiation. Give examples of each.
 Conduction is a transfer of energy by contact. Convection is a transfer of energy by movement of fluid. Radiation is a transfer of energy by electromagnetic waves.

2. Explain the following phase change diagram.



3. Students will determine relationships among force, mass and motion.
 - a. Explain the difference between distance and displacement.
 Distance is the total path length. Displacement is the change in position.
 - b. What is the difference between average speed and instantaneous speed?
 Average speed is total distance divided by total time. Instantaneous speed is speed at a particular time.
 - c. What is the formula for average speed?

$$v = \frac{d}{t}$$

$$v = \frac{d}{t}$$

Physical Science Eoct Study Guide Energy Transformations

KJ Lindholm-Leary



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