

## Chapter 6 Chemical Bonds

**Section 6.2 Covalent Bonding****(pages 165–169)**

*This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.*

**Reading Strategy (page 165)**

**Relating Text and Visuals** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

**Covalent Bonds (pages 165–167)**

- Describe a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule. \_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - zero
  - one
  - two
  - three

# Physical Science Chapter 6 Chemical Bonds Chapter Review

**Robert B. Northrop, Anne N. Connor**



## **Physical Science Chapter 6 Chemical Bonds Chapter Review:**

## Whispering the Secrets of Language: An Psychological Journey through **Physical Science Chapter 6 Chemical Bonds Chapter Review**

In a digitally-driven world wherever displays reign great and immediate communication drowns out the subtleties of language, the profound techniques and emotional subtleties concealed within phrases often move unheard. Yet, located within the pages of **Physical Science Chapter 6 Chemical Bonds Chapter Review** a captivating fictional treasure blinking with organic emotions, lies an exceptional quest waiting to be undertaken. Composed by a skilled wordsmith, that enchanting opus encourages visitors on an introspective journey, gently unraveling the veiled truths and profound influence resonating within ab muscles cloth of every word. Within the emotional depths of this touching review, we can embark upon a heartfelt exploration of the book is key styles, dissect its captivating publishing model, and succumb to the powerful resonance it evokes strong within the recesses of readers hearts.

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