

Bonding and Reactions

Study Guide

10 pages including KEY

Chemical Bonding and Reactions Study Guide

This study packet will help you study for the Chemical Bonding and Reactions exam by highlighting the most important concepts on the exam. This test will evaluate your understanding of the material you have learned for your first term of the chemical and physics courses.

B. VOCABULARY

Atom	Electron	Electron transfer	Electron transfer
Atomic mass	Electron	Electron transfer	Electron transfer
Atomic weight	Electron	Electron transfer	Electron transfer
Atomic	Electron	Electron transfer	Electron transfer
Chemical bond	Electron	Electron transfer	Electron transfer
Chemical reaction	Electron	Electron transfer	Electron transfer

C. READING

1. Why do atoms bond together?
2. What is a covalent bond?
3. What kind of electron flow is covalent bond?
4. What is an ionic bond?
5. What kind of electron flow is ionic bond?
6. Why don't noble gases bond together with other elements?

Write the formula for the following compounds with multiple valence numbers.

1. Copper (II) chloride _____
2. Copper (II) chloride _____

Write the following compounds with multiple valence numbers.

1. Iron _____
2. Iron _____

REACTIONS

What happens in an oxidation-reduction reaction?

What does an oxidation-reduction reaction involve?

What is an oxidation-reduction reaction?

What is an oxidation-reduction reaction?

What is the law of conservation of mass?

What is the law of conservation of mass?

A coefficient in a chemical equation is a number placed in front of a chemical formula to indicate the number of molecules of that substance that are involved in the reaction.

What is a chemical reaction? A chemical reaction is a process in which one or more substances are converted into one or more different substances.

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B. VOCABULARY

1. Atom: A negatively charged particle that is present in all matter.
2. Atomic mass: The mass of an atom, measured in atomic mass units (amu).
3. Atomic weight: The average atomic mass of an element, measured in atomic mass units (amu).
4. Chemical bond: A force that holds atoms together in a molecule or compound.
5. Chemical reaction: A process in which one or more substances are converted into one or more different substances.
6. Coefficient: A number placed in front of a chemical formula to indicate the number of molecules of that substance that are involved in the reaction.
7. Electron: A negatively charged particle that is present in all matter.
8. Electron transfer: A process in which one or more electrons are transferred from one atom to another.
9. Electronegativity: A measure of an atom's ability to attract electrons in a chemical bond.
10. Electropositivity: A measure of an atom's ability to lose electrons in a chemical reaction.
11. Electropositive: A measure of an atom's ability to lose electrons in a chemical reaction.
12. Electropositive: A measure of an atom's ability to lose electrons in a chemical reaction.
13. Electropositive: A measure of an atom's ability to lose electrons in a chemical reaction.
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Bonding and Chemical Reactions - KEY

C. READING

1. Why do atoms bond together? Atoms bond together to achieve a more stable electronic configuration, typically by filling or emptying their valence electron shells to satisfy the octet rule.
2. What is a covalent bond? A covalent bond is a type of chemical bond formed between two atoms by the sharing of electron pairs. It is typically found between non-metal atoms.
3. What is an ionic bond? An ionic bond is a type of chemical bond formed between two atoms by the transfer of electrons. It is typically found between a metal atom and a non-metal atom.
4. What is a metallic bond? A metallic bond is a type of chemical bond found in metals, where the valence electrons are delocalized and form a "sea of electrons" that holds the metal cations together.
5. Why don't noble gases bond together with other elements? Noble gases have a full valence shell, which makes them very stable and unreactive. They do not have a strong tendency to gain or lose electrons to form a bond.
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