

BAGONG SILANG NATIONAL HIGH SCHOOL
1ST QUARTER EXAMINATION
PHYSICAL SCIENCE

NAME: _____
GRADE: _____

DATE: _____
SCORE: _____

Directions: Write the letter of the correct answer on the space provided.

- _____ 1. It is also known as nuclear fusion and the formation of new nuclei actions in the early stages of development of the universe.
a. Nucleosynthesis b. S-Process c. R-Process d. Singularity
- _____ 2-4. There are two categories of star, the average star and massive star. The life cycle of these two stars are common from 1st to 3rd stages namely nebula, protostar, and main sequence star. What will be the next stages in the life of an average star until it reaches its death?
a. Red super giant → Supernova → Black hole → neutron star 0
b. Red giant → white dwarf → black dwarf 3
c. Red giant → white dwarf → black hole 2
d. Red supergiant → white dwarf → black dwarf 1
- _____ 5. He synthesized element with atomic number 43 using a linear particle accelerator.
a. Ernest Rutherford b. Ernest Lawrence c. Dmitri Mendeleev d. John Dalton
- _____ 6. The following are light elements EXCEPT:
a. helium b. hydrogen c. lithium d. iron
- _____ 7. Which of the following molecule is nonpolar?
a. NaCl b. HCl c. CO₂ d. NH₃
- _____ 8. What type of chemical bond holds the atoms of water molecule together?
a. Hydrogen bond b. Ionic bond c. Polar covalent bond d. Nonpolar covalent bond
- _____ 9. Which of the following IMFAs is considered as the weakest?
a. H-bonding b. Ion-dipole c. Dipole-dipole d. London forces
- _____ 10. Which of the following statements is TRUE about nonpolar molecules?
a. high boiling point c. low surface tension
b. high melting point d. low vapor pressure
- _____ 11. Forces that are present between an ion and a polar molecule
a. ion-dipole b. dipole-dipole c. dispersion forces d. hydrogen bonding
- _____ 12. H-bonding forms when the substances involved are polar and there are _____.
a. H-atoms of a molecule interact with to O, N, F of other molecules.
b. C-atoms attached to O, N, F.
c. central atoms with O, N, F as attached atoms.
d. unshared pair of electrons in the central atom.
- _____ 13. The amount of energy required to stretch or increase the surface of a liquid by a unit area
a. specific heat b. surface tension c. vapor pressure d. heat of vaporization
- _____ 14. Liquids can form spherical elastic film to minimize surface area. What intermolecular force is responsible for the formation of this film in water?
a. H-bonding b. ion-induced dipole c. dipole-induced dipole d. London dispersion force
- _____ 15. Which group of biomolecules carries and passes on the hereditary information of the organisms?
a. carbohydrates b. lipids c. nucleic acids d. proteins
- _____ 16. What are the elements that make up carbohydrates?
a. C and H b. C, H and N c. C, H and O d. C, H, O and N
- _____ 17-19. Is carbohydrates one of the most important biological macromolecules needed by a human body?
a. Yes, because it is the primary source of energy. 3
b. Yes, because it can be a storage of energy for a long period of time. 1
c. Yes, because it can be found in rice, cereals, fruits and pastas. 2
d. Yes, because it carries genetic information. 0
- _____ 20. The measure of how fast is the change in the concentration of the reactants or products in a chemical reaction.
a. Activation Energy b. Activation Rate c. Chemical Reaction d. Concentration
- _____ 21. A chemical change that occurs when two or more substances combine to form a new substance.
a. Activation Energy b. Chemical Kinetics c. Chemical Reaction d. Product

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1. Understanding the eBook Physical Science Answer Key On Ibooks
 - The Rise of Digital Reading Physical Science Answer Key On Ibooks
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Answer Key On Ibooks
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Physical Science Answer Key On Ibooks
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Answer Key On Ibooks
 - Personalized Recommendations

- Physical Science Answer Key On Ibooks User Reviews and Ratings
- Physical Science Answer Key On Ibooks and Bestseller Lists
- 5. Accessing Physical Science Answer Key On Ibooks Free and Paid eBooks
 - Physical Science Answer Key On Ibooks Public Domain eBooks
 - Physical Science Answer Key On Ibooks eBook Subscription Services
 - Physical Science Answer Key On Ibooks Budget-Friendly Options
- 6. Navigating Physical Science Answer Key On Ibooks eBook Formats
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 - Adjustable Fonts and Text Sizes of Physical Science Answer Key On Ibooks
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 - Interactive Elements Physical Science Answer Key On Ibooks
- 8. Staying Engaged with Physical Science Answer Key On Ibooks
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Answer Key On Ibooks
- 9. Balancing eBooks and Physical Books Physical Science Answer Key On Ibooks
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Answer Key On Ibooks
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Answer Key On Ibooks
 - Setting Reading Goals Physical Science Answer Key On Ibooks
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Answer Key On Ibooks
 - Fact-Checking eBook Content of Physical Science Answer Key On Ibooks

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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