



MERCY JUNIOR COLLEGE
Grade 11
Physical Science
Midterm Exam

NAME: _____
YEAR LEVEL & SECTION: _____
TEACHER'S NAME: _____
PROCTOR'S NAME: _____

I. MULTIPLE CHOICE

INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)

- It is a subatomic particle, symbol e^- or β^- , with a negative elementary electric charge.
A. Neutron
B. Electron
C. Proton
D. Atom
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
A. Neutron
B. Electron
C. Proton
D. Atom
- It is a subatomic particle, symbol p or p^+ , with a positive electric charge of $+1e$ elementary charge and mass slightly less than that of a neutron.
A. Neutron
B. Electron
C. Proton
D. Atom
- It is a subatomic particle, symbol n or n^0 , with no net electric charge and a mass slightly larger than that of a proton.
A. Neutron
B. Electron
C. Proton
D. Atom
- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- It is a type of radioactive decay in which an atomic nucleus emits an alpha particle and thereby transforms or 'decays' into an atom with a mass number that is reduced by four and an atomic number that is reduced by two.
A. Alpha Decay
B. Beta Decay
C. Gamma Ray
D. Radiation

- It is a type of radioactive decay in which a beta ray (fast energetic electron or positron) and a neutrino are emitted from an atomic nucleus.

- Alpha Decay
- Beta Decay
- Gamma Ray
- Radiation

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.

- Electron Configuration
- Periodic Table of Elements
- Orbital Models
- Energy Levels

- What type of nuclear reaction is this?



- Alpha Decay
- Beta Decay
- Gamma Ray
- No nuclear reaction

Physical Science 2june Exam Paper 1

Rachel S Tattersall



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Table of Contents Physical Science 2june Exam Paper 1

1. Understanding the eBook Physical Science 2june Exam Paper 1
 - The Rise of Digital Reading Physical Science 2june Exam Paper 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science 2june Exam Paper 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science 2june Exam Paper 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science 2june Exam Paper 1
 - Personalized Recommendations
 - Physical Science 2june Exam Paper 1 User Reviews and Ratings
 - Physical Science 2june Exam Paper 1 and Bestseller Lists
5. Accessing Physical Science 2june Exam Paper 1 Free and Paid eBooks
 - Physical Science 2june Exam Paper 1 Public Domain eBooks
 - Physical Science 2june Exam Paper 1 eBook Subscription Services
 - Physical Science 2june Exam Paper 1 Budget-Friendly Options

6. Navigating Physical Science 2june Exam Paper 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science 2june Exam Paper 1 Compatibility with Devices
 - Physical Science 2june Exam Paper 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science 2june Exam Paper 1
 - Highlighting and Note-Taking Physical Science 2june Exam Paper 1
 - Interactive Elements Physical Science 2june Exam Paper 1
8. Staying Engaged with Physical Science 2june Exam Paper 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science 2june Exam Paper 1
9. Balancing eBooks and Physical Books Physical Science 2june Exam Paper 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science 2june Exam Paper 1
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science 2june Exam Paper 1
 - Setting Reading Goals Physical Science 2june Exam Paper 1
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science 2june Exam Paper 1
 - Fact-Checking eBook Content of Physical Science 2june Exam Paper 1
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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