

SOLUBILITY (POLAR VS. NONPOLAR)

Name _____

Generally, "like dissolves like." Polar molecules dissolve other polar molecules and ionic compounds. Nonpolar molecules dissolve other nonpolar molecules. Alcohols, which have characteristics of both, tend to dissolve in both types of solvents, but will not dissolve ionic solids.

Check the appropriate columns as to whether the solute is soluble in a polar or nonpolar solvent.

SOLUTES	SOLVENTS		
	Water	CCl_4	Alcohol
1. NaCl	/		
2. I_2		/	
3. ethanol			
4. benzene C_6H_6			
5. Br_2			
6. KNO_3			
7. toluene C_7H_8			
8. Ca(OH)_2			

Polar Vs Nonpolar Instructional Fair

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Polar Vs Nonpolar Instructional Fair:

Polar Vs Nonpolar Instructional Fair Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Polar Vs Nonpolar Instructional Fair**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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