Chemistry pH and pOH Worksheet

1. Calculate the values of both pH and pOH of the following solutions.

	pН	рОН
0.020 M HCI		
0.0050 M NaOH		
A blood sample 7.2 x 10° M of H		
0.0050 M NaOH		

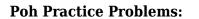
2. Find the values of [H¹], pOH, and [OH¹] that correspond to each of the following pH values.

	[H+]	[OH-]	рОН
pH of lemon juice = 2.9			
pH of sauerkraut = 3.85			
pH of milk of magnesia = 10.81			
pH of orange juice – 4.11			
pH of diluted household ammonia = 11,61			

- 3. Determine which of the solutions in problem #2 are acidic.
- 4. A certain brand of root beer has a hydrogen concentration equal to 1.9×10^{-5} M. What are the pH and pOH of this root beer?
- 5. Dr. Pepper has a $[H^*] = 1.4 \times 10^{-5} M$. What is its pH?

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