

Key

Molecular Geometry

How can molecular shapes be predicted using the VSEPR theory?

Why?

When you draw a Lewis structure for a molecule on paper, you are making a two-dimensional representation of the atoms. In reality however, molecules are not flat—they are *three-dimensional*. The true shape of a molecule is important because it determines many physical and chemical properties for the substance. In this activity you will learn how to predict molecular shapes.

Model 1 – Lewis Structures

Lewis Structures		3-D Molecular Shape
1. H_2CO	3 electron domains (3 bonding, 0 nonbonding)	trigonal planar
2. BeF_2	2 electron domains (2 bonding, 0 nonbonding)	Linear
3. CH_4	4 electron domains (4 bonding, 0 nonbonding)	Tetrahedral
4. NH_3	4 electron domains (3 bonding, 1 nonbonding)	pyramidal
5. H_2O	4 electron domains (2 bonding, 2 nonbonding)	Bent
6. CO_2	2 electron domains (2 bonding, 0 nonbonding)	linear

Lone pair = ..

Pogil Molecular Geometry Answers

Huangqi Zhang

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Pogil Molecular Geometry Answers:

Pogil Molecular Geometry Answers Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the energy of words has be more evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Pogil Molecular Geometry Answers**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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