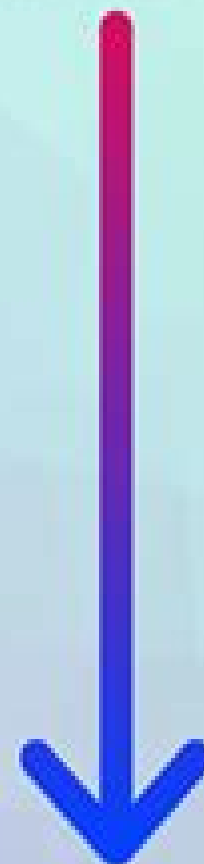


Reactivity Series of Metals

Lithium	$\text{Li} \rightleftharpoons \text{Li}^+ + \text{e}^-$
Barium	$\text{Ba} \rightleftharpoons \text{Ba}^{2+} + 2\text{e}^-$
Calcium	$\text{Ca} \rightleftharpoons \text{Ca}^{2+} + 2\text{e}^-$
Magnesium	$\text{Mg} \rightleftharpoons \text{Mg}^{2+} + 2\text{e}^-$
Aluminum	$\text{Al} \rightleftharpoons \text{Al}^{3+} + 3\text{e}^-$
Manganese	$\text{Mn} \rightleftharpoons \text{Mn}^{2+} + 2\text{e}^-$
Zinc	$\text{Zn} \rightleftharpoons \text{Zn}^{2+} + 2\text{e}^-$
Chromium	$\text{Cr} \rightleftharpoons \text{Cr}^{3+} + 3\text{e}^-$
Iron	$\text{Fe} \rightleftharpoons \text{Fe}^{2+} + 2\text{e}^-$
Cobalt	$\text{Co} \rightleftharpoons \text{Co}^{2+} + 2\text{e}^-$
Nickel	$\text{Ni} \rightleftharpoons \text{Ni}^{2+} + 2\text{e}^-$
Tin	$\text{Sn} \rightleftharpoons \text{Sn}^{2+} + 2\text{e}^-$
Lead	$\text{Pb} \rightleftharpoons \text{Pb}^{2+} + 2\text{e}^-$
Hydrogen	$\text{H} \rightleftharpoons 2\text{H}^+ + 2\text{e}^-$
Copper	$\text{Cu} \rightleftharpoons \text{Cu}^{2+} + 2\text{e}^-$
Silver	$\text{Ag} \rightleftharpoons \text{Ag}^+ + \text{e}^-$
Platinum	$\text{Pt} \rightleftharpoons \text{Pt}^{2+} + 2\text{e}^-$
Gold	$\text{Au} \rightleftharpoons \text{Au}^{3+} + 3\text{e}^-$

most reactive



least reactive

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