

Basic Solubility Rules (View 3)

General Solubility Rules (for water at 25°C)

<u>Ion</u>	<u>Solubility</u>	<u>Exceptions</u>
Alkali ions	soluble	none
NH_4^{1+} , H^{1+}	soluble	none
Cl^{1-} , I^{1-} , Br^{1-}	soluble	except Ag^+ , Hg_2^{2+} , Pb^{2+}
$\text{CH}_3\text{COO}^{1-}$, NO_3^{1-}	soluble	none
OH^{1-}	insoluble	except Alkali ions, Ca^{2+} , Ba^{2+} , Sr^{2+}
ClO_4^{1-}	soluble	none
SO_4^{2-}	soluble	except Ca^{2+} , Ba^{2+} , Sr^{2+} , Hg^{2+} , Pb^{2+} , Ag^+
CO_3^{2-} , PO_4^{3-} , CrO_4^{2-} , S^{2-} , O^{2-}	insoluble	except Alkali ions and NH_4^+

Use this one – found on the back of our Periodic Table

Pogil Solubility Rules Answers

LL Leslie



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