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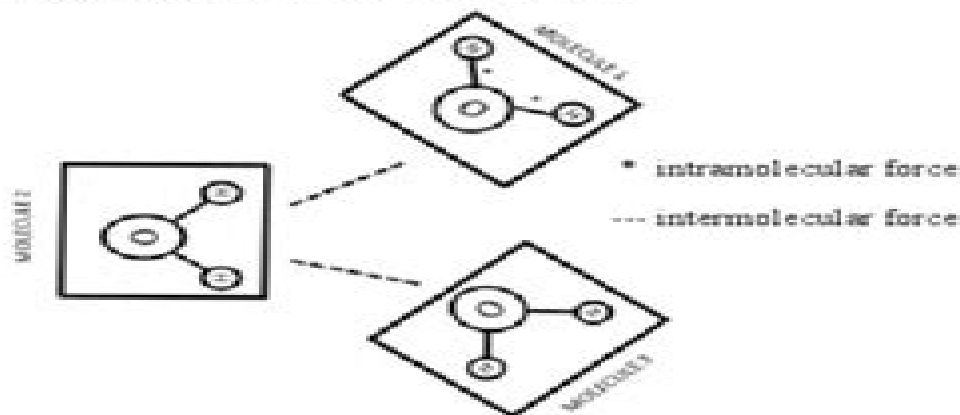
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POGIL: Intermolecular Forces

Model 1: What is an intermolecular force?

As you have learned, matter is made up of discrete particles called **atoms**, which chemically combine to form **molecules**. Molecules do not exist as independent units; in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are **intermolecular forces**. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intramolecular and Intermolecular Forces



Critical Thinking Questions:

1. What specific molecule is represented inside each box in Figure 1?
2. In relation to the box for molecule 1, where do the intramolecular forces exist in Figure 1 – inside the boxes or outside the boxes?
3. Based on the intramolecular forces for molecule 1, draw similar asterisks (*) for the intramolecular forces on the diagram for molecules 2 and 3.
4. In relation to the molecule, where do intramolecular forces tend to occur – within the molecule or outside of the molecule?
5. Two intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules – within the molecules or between the molecules?
6. State the difference between intermolecular and intramolecular forces in terms of where they occur on the molecular level.



Pogil Intermolecular Forces

Christian Drosten



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