

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in chemical gases present in the air. Metals react or combine in the presence of air and water. Nitrate(II) ions decompose to give the reaction for decomposition with the use of nitric acid as product from nitrate salt compounds. All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this activity, you will explore when is happening on the atomic level in redox reactions.

Model 1 – Redox Reactions

Redox Reactions:



Decomposition Reactions:



1. What are typical reactions you observe in Model 1?

Redox reactions and decomposition reactions.

2. Explain the redox and decomposition reactions in Model 1. What does decomposition in the redox reactions also include other products differently than in redox reactions? If yes, what specific examples from Model 1 are appropriate your answer.

There is no reaction showing decomposition reaction in Model 1.

Pogil Chemistry Answer Key Oxidation And Activities

Albert A Gayle



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