

16. The dissociation of water is shown below:



- At 25 °C, the auto product of water,  $K_w$ , is  $1.0 \times 10^{-14}$  mol/L.  
At 65 °C, the auto product of water,  $K_w$ , is  $1.0 \times 10^{-13}$  mol/L.

17. Explain whether the dissociation of water is an endothermic or exothermic process.  
Endothermic

18. Predict, using a calculation, whether a solution of 1.0 M  $HCl$  is neutral, acidic or basic. (2)

Acidic

19. Explain by calculating pH.

20.  $pK_a$ ,  $pH$  and  $pOH$  are significant notions.

Calculate  $pK_a$  of  $HCl$ .

Give your answer to two decimal places.

$$-\log(4.31 \times 10^{-14}) = 13.07$$

$$pK_a = 13.03 \quad (1)$$

21. A solution of hydrochloric acid is mixed with water and the resulting solution is neutral at 25 °C.

Calculate the pH of the resulting solution at 65 °C.

Give your answer to two decimal places.

$$pH = 13.07$$

pH = \_\_\_\_\_ (2)  
(Total: 10)

# Ocr Chemistry A2 Unofficial Mark Scheme 2014

**Pasan Witharana**



## **Ocr Chemistry A2 Unofficial Mark Scheme 2014:**

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OCR(B) AS/A2 Chemistry (Salters) Student Unit Guide: Units F333 and F336 Chemistry in Practice and Individual Investigation Frank Harriss, 2011-12-09 Student Unit Guides are perfect for revision Each guide is written by an examiner and explains the unit requirements This guide offers advice on preparing for the OCR B Salters Units F333 Chemistry in Practice and F336 Individual Investigation assessments The Practical Chemistry section deals with experimental results and practical techniques It explains how to obtain and record results and how to process them

*AS/A2 Level Chemistry OCR A Complete Revision & Practice* Richard Parsons, 2010-01-01 This revision guide for A Level Chemistry is suitable for the OCR A exam board Tricky topics are broken down and explained in straightforward terms with plenty of diagrams The book also includes practice exam questions for every topic

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- Chapter 5: Conclusion

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