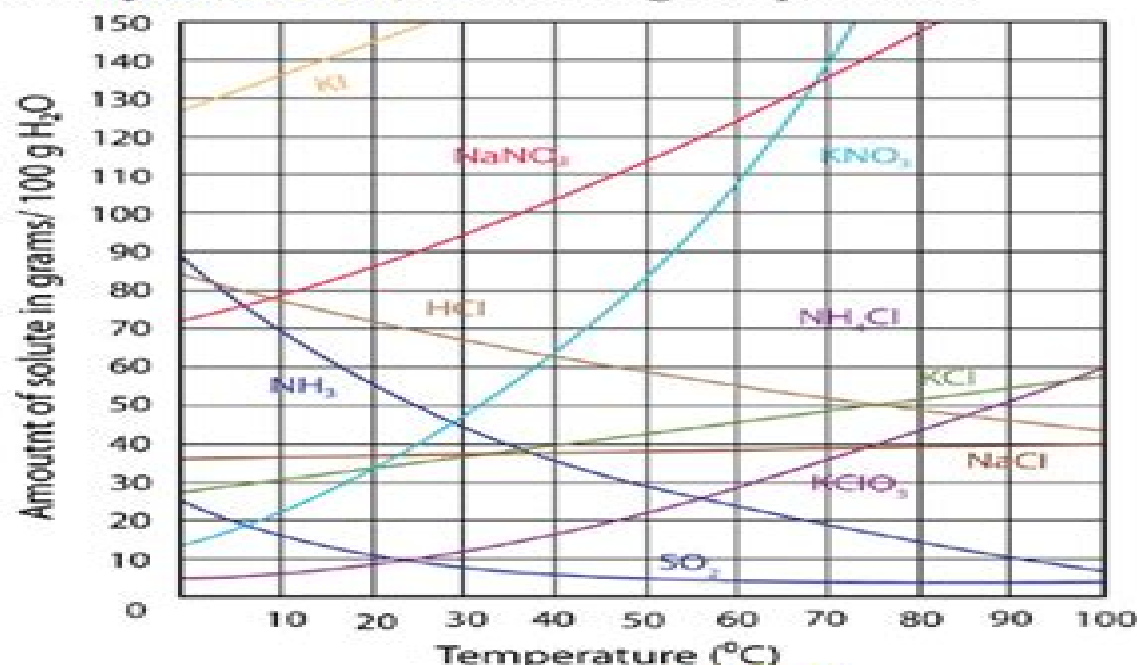




Interpreting Solubility Curve



1. Define solubility. The maximum amount of solute that can dissolve in a solvent
2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve? H₂O
- II. Saturation points are given per how much of the solvent? 10
- III. What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
 10°C = 79 40°C = 103 80°C = 147
- IV. How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
 10°C = 70 30°C = 45 90°C = 10
- V. Which salt is the least soluble in water? NaCl
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C? _____
- VII. At 40 °C, how much potassium nitrate will be dissolved in 300 g of water? _____
- VIII. Which salt shows the least change in solubility from 0° to 100° C? _____
- IX. Which substance shows a decrease in solubility from 0° to 100° C? _____
- X. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? _____

Pogil Interpreting Solubility Curves Answer Key

D Kirk

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Pogil Interpreting Solubility Curves Answer Key:

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Pogil Interpreting Solubility Curves Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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