

POGIL Activity – Gas Variables

(Teacher Notes)

- Assessment Questions and Responses:

Assessment Questions

1. Select the equation below that gives the correct relationship between the internal pressure of a gas and the volume of a flexible container.
 - a. $P = k/r$
 - b. $P = kV$
 - c. $V = kP$
 - d. $k = p/V$
2. Explain, on the molecular level, the change in internal pressure of a gas sample in a nonflexible container when the temperature is raised.
3. Explain, on the molecular level, the change in volume in a flexible container when more gas is added.

Assessment Target Responses

1. a.
2. *When the temperature of a gas is increased, the average kinetic energy of the molecules increases. This makes the gas molecules hit harder and more often, which increases the pressure.*

Pogil Gas Variables

J Spring



Pogil Gas Variables:

Adopting the Melody of Expression: An Psychological Symphony within **Pogil Gas Variables**

In a global consumed by displays and the ceaseless chatter of quick interaction, the melodic beauty and emotional symphony created by the prepared word frequently disappear in to the back ground, eclipsed by the relentless sound and disruptions that permeate our lives. Nevertheless, nestled within the pages of **Pogil Gas Variables** an enchanting literary treasure full of natural thoughts, lies an immersive symphony waiting to be embraced. Crafted by a masterful musician of language, that captivating masterpiece conducts readers on a psychological trip, skillfully unraveling the hidden melodies and profound influence resonating within each cautiously constructed phrase. Within the depths with this moving evaluation, we shall examine the book is main harmonies, analyze their enthralling writing style, and submit ourselves to the profound resonance that echoes in the depths of readers souls.

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