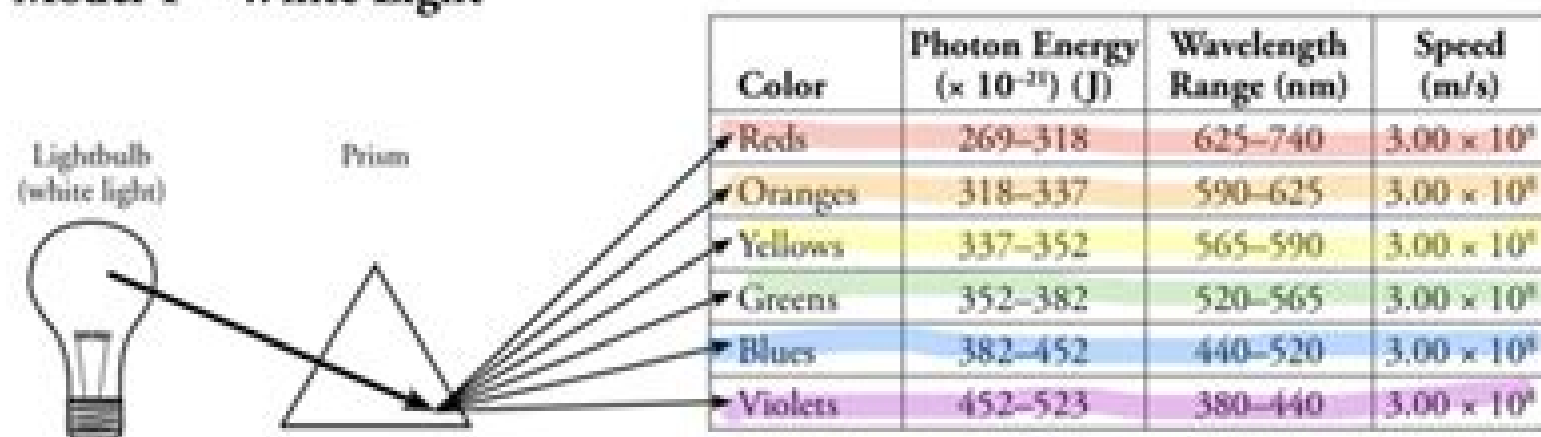


Trace materials, such as evidence from a crime scene, lead in paint or mercury in drinking water, can be identified by heating or burning the materials and examining the color(s) of light given off in the form of bright-line spectra.

Model 1 – White Light



1. Trace the arrows in Model 1 and shade in the table with colored pencils where appropriate.
2. What happens to white light when it passes through a prism?

it splits into a spectrum of colors

3. Why are the color labels in the table in Model 1 plural (*i.e.*, "Reds" rather than "Red")?

because there are multiple shades of each color

4. Do all colors of light travel at the same speed?

yes the speed of light is the same no matter the color

5. Do all colors of light have the same energy? If no, which colors have the highest energy and the least energy, respectively?

warm colors have less energy cool colors have more

Pogil Electron Energy And Light Answer Key

Rosina Ehmann



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The Energy of Photo-electrons from Sodium and Potassium as a Function of the Frequency of the Incident Light William Henry Kadesch,1914 **Energy: Light Energy** George Graybill,2013-10-01 This is the chapter slice Light Energy from the full lesson plan Energy Unlock the mysteries of energy Energy is more than the ability to do work we present these concepts in a way that makes them more accessible to students and easier to understand The best way to understand energy is to first look at all the different kinds of energy including What Is Energy Mechanical Energy Thermal Sound Energy and Waves as well as Light Energy Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary We also explore other forms of potential energy as well as how energy moves and changes Written to grade and comprised of reading passages student activities and color mini posters our resource can be used effectively for your whole class All of our content meets the Common Core State Standards and are written to Bloom s Taxonomy and STEM initiatives *Photo-electricity, the Liberation of Electrons by Light. With Chapters on Fluorescence & Phosphorescence, and Photo-chemical Actions & Photography* Herbert Stanley Allen,1913 *Energy and Light* Peter Lafferty,1989 **Light Bulbs, Switches and Batteries** Heron Books,2018-11-22 In this illustrated book students are introduced to the concepts of energy electrons and electrical energy They learn about voltage current and resistance and Ohm s law which relates to all three Students learn where the energy to light the light bulbs comes from and how it can be turned on and off The book includes activities which direct students to connect light bulbs to batteries with wires and experiment to find out what happens when they use more batteries or more light bulbs or different kinds of wire The accompanying Learning Guide directs the student through the book and its activities and also provides additional activities that assist the student in fully understanding and applying the concepts learned Photo Electricity Herbert Stanley Allen,Frank Horton,Joseph John Thomson,1913

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