

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in chemical gases present in the air. Metals react or combine in the presence of air and water. Nitrate(II) ions decomposed through the reaction to decomposed into the gas of nitrogen as peroxide gas. Metals and compounds, PDB of these reactions are examples of oxidation and reduction, reduction happens in metal reactions. In this reaction, you will replace what is happening in the reaction level in metal reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are typical reactions in Model 1?

Redox reactions and reduction reactions.

2. Explain the redox and reduction reactions in Model 1. What does copper(II) nitrate do in the redox reactions? What would other gases do in the redox reactions? It gets oxidized and reduced. Since Model 1 is copper(II) nitrate.

There is no reaction showing the redox reactions in the redox reactions.

Pogil Chemistry Answer Key Oxidation

Joacim Rocklöv



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Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Pogil Chemistry Answer Key Oxidation**

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