- 1. Which illustration below represents
 - a. solute particles in a solid state in water?
 - b. solute particles in an aqueous state?



2. What variables are controlled in all five beakers of Model 1?

Same amount of solvent(water), same temperature, same solute, stirred for 2 hours

Count the particles present in each beaker of Model 1. Fill in the table to show the number of dissolved solute particles and the number of solid solute particles.

Refer to table.

- 4. Consider the beakers in Model 1.
 - a. Which beakers represent unsaturated solutions?
 - A AL BU
 - b. Which beakers represent saturated solutions?
 - C. D&E
- 5. Beakers A—E in Model I are depicted as representing five different or separate solutions. They could also be considered as five "snapshots" of the same beaker over time. In other words, if additional measured quantities of solute were stirred into beaker A in small increments over time, then beakers B—E would result.
 - a. When a small amount of additional solute is added to an unsaturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.
 - The particles dissolve. If we look at beaker A and we add more solute it becomes like beaker B which has more particles dissolved.
 - b. When a small amount of additional solute is added to a saturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.

The particles settle out to the bottom, this is like going from beaker C to D.

c. Predict what would happen if a small amount of additional solute were stirred into beaker E in Model 1.

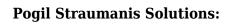
Nothing would happen, more solute would settle to the bottom.

Have each person in your group provide an example of the word "saturated" as it is used in an everyday context. Summarize the meaning of the word in the space below.

Some examples: a soaked sponge, soil after a heavy rain, adding sugar to iced tea Definition: something that has the maximum amount of a substance in it.

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Michael Brown



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Table of Contents Pogil Straumanis Solutions

- 1. Understanding the eBook Pogil Straumanis Solutions
 - The Rise of Digital Reading Pogil Straumanis Solutions
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Pogil Straumanis Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Straumanis Solutions
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Pogil Straumanis Solutions
 - Personalized Recommendations
 - Pogil Straumanis Solutions User Reviews and Ratings
 - Pogil Straumanis Solutions and Bestseller Lists
- 5. Accessing Pogil Straumanis Solutions Free and Paid eBooks
 - Pogil Straumanis Solutions Public Domain eBooks
 - Pogil Straumanis Solutions eBook Subscription Services
 - Pogil Straumanis Solutions Budget-Friendly Options
- 6. Navigating Pogil Straumanis Solutions eBook Formats
 - o ePub, PDF, MOBI, and More
 - Pogil Straumanis Solutions Compatibility with Devices
 - Pogil Straumanis Solutions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Straumanis Solutions
 - Highlighting and Note-Taking Pogil Straumanis Solutions
 - Interactive Elements Pogil Straumanis Solutions
- 8. Staying Engaged with Pogil Straumanis Solutions

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Pogil Straumanis Solutions
- 9. Balancing eBooks and Physical Books Pogil Straumanis Solutions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Straumanis Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil Straumanis Solutions
 - Setting Reading Goals Pogil Straumanis Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil Straumanis Solutions
 - Fact-Checking eBook Content of Pogil Straumanis Solutions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - $\circ \ \ Integration \ of \ Multimedia \ Elements$
 - Interactive and Gamified eBooks

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