Model 2 - Prefixes and Suffixes

Prefix	Numerical Value
maomo-	1
di-	2
tci-	3
ectra-	4
penta-	5
hexa-	6
Берка-	7
octa-	8
monta-	9
deca-	10

Molecular Formula	Name of Compound
BCI,	Boron trichloride
SF _e	Solfur hexafluoride
IF.	Iodine heptafluoride
NI,	Nitrogen triiodide
N,O,	Dinitrogen tetroxide
CI,O	Dichlorine monoxide
P _i O _{in}	Tetraphosphorus decoxide
B,H,	Pentaboron nonahydride
Br,O,	Tribromine octoxide
CIF	Chlorine monofluoride

- 5. Examine the prefixes in Model 2. Hill in the numerical value that corresponds to each prefix.
- 6. What suffix (ending) do all the compound names in Model 2 have in common? They all end in "ide"



- 7. Carefully examine the names of the compounds in Model 2. When is a prefix NOT used in from of the name of an element of the name of an element of the first element is one. (Examples: IFn-locline high flow the Wig-nitrogen triid ide
 - Consider the compound NO.
 - a. Which element, nitrogen or oxygen, would require a prefix in the molecule name? Explain

Oxygen would require a prefix of "mono" because it is the second element and there is one oxygen. Nitrogen does not require a prefix because it is the first element and there is only a Name the molecule NO. nitroser.



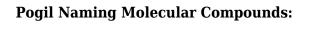
nitrogen monoxide (Notice, the "o" of minous diget 9. Find two compounds in Model 2 that contain a subscript of "4" in their molecular formula.

- - N2 O4 dinitrogen tetroxide II P4 O10-Tetrophosphorus decexide a. List the formulas and names for the two compounds.
 - b. What is different about the spelling of the prefix meaning "four" in these two names? The "a" has been dropped from the end of the profix tetra.

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