

Oxidation and Reduction

When happens when elements are transformed in a chemical reaction?

What?

Below reactions show us process in nature as it occurs with either compounds in the air. Copper gets oxidized in dissolved gases present in lakes. Metals react or corrode in the presence of dissolved water. Sulfuric acid is transported through the air from the atmosphere and then rain or other water in presence of sulfuric acid compounds. All of these reactions are examples of oxidation and reduction, otherwise known as redox reactions. In this section you will explore what is happening in the atoms level in redox reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



- What are typical reactions you observe in Model 1?

Redox reactions and redox reactions.

- Examine the redox and reduction reactions in Model 1. What does ΔH_{redox} represent in the redox reactions? How would you determine ΔH_{redox} if you had specific information from Model 1 on specific redox reactions? There is no obvious straightforward feature that can be used to determine ΔH_{redox} .

Pogil Lab Oxidation Answer Key

Raffaela Di Napoli

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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