

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- At standard atmospheric conditions, all gases have approximately the same kinetic energy, and are made up of molecules that behave as independent particles (the postulates of the kinetic molecular theory).
- At standard atmospheric conditions, all molecules have the same average kinetic energy.
- At standard atmospheric conditions, there is equal force exerted by all molecules on a surface.

Model 1: Representation of Gases in Different Phases

The following figure illustrates the representation of particles in three different phases of matter.



Figure 1: Representation of particles in different phases of matter

Model 1 represents the particles in different phases of matter. The particles in the gas phase are widely spaced and move in all directions. The particles in the liquid phase are closely packed and move in all directions. The particles in the crystalline solid phase are tightly packed and move in all directions.

Key Questions

1. Which one of the three representations of particles best represents the gas phase? Which one of the three representations of particles best represents the liquid phase? Which one of the three representations of particles best represents the crystalline solid phase? Justify your answers.

Table 1: Representation of particles in different phases of matter

	Gas	Liquid	Crystalline Solid
Which one of the three representations of particles best represents the gas phase?	Model 1: Gas. The particles are widely spaced and move in all directions. This is the best representation of the gas phase.	Model 2: Liquid. The particles are closely packed and move in all directions. This is the best representation of the liquid phase.	Model 3: Crystalline Solid. The particles are tightly packed and move in all directions. This is the best representation of the crystalline solid phase.
Which one of the three representations of particles best represents the liquid phase?	Model 2: Liquid. The particles are closely packed and move in all directions. This is the best representation of the liquid phase.	Model 1: Gas. The particles are widely spaced and move in all directions. This is the best representation of the gas phase.	Model 3: Crystalline Solid. The particles are tightly packed and move in all directions. This is the best representation of the crystalline solid phase.
Which one of the three representations of particles best represents the crystalline solid phase?	Model 3: Crystalline Solid. The particles are tightly packed and move in all directions. This is the best representation of the crystalline solid phase.	Model 2: Liquid. The particles are closely packed and move in all directions. This is the best representation of the liquid phase.	Model 1: Gas. The particles are widely spaced and move in all directions. This is the best representation of the gas phase.

Pogil Kinetic Molecular Answers

Keith S Taber



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Chemical Pedagogy Keith S Taber, 2024-12-20 How should chemistry be taught in schools colleges and universities Chemical Pedagogy discusses teaching approaches and techniques the reasoning behind them and the evidence for their effectiveness The book surveys a wide range of different pedagogic strategies and tactics that have been recommended to better engage learners and provide more effective chemistry teaching These accounts are supported by an initial introduction to some key ideas and debates about pedagogy the science of teaching Chemical Pedagogy discusses how teaching innovations can be tested to inform research based practice Through this book the author explores the challenges of carrying out valid experimental studies in education and the impediments to generalising study results to diverse teaching and learning contexts As a result the author highlights both the need to read published studies critically and the value of teachers and lecturers testing out recommended innovations in their own classrooms Chemical Pedagogy introduces core principles from research into human cognition and learning to provide a theoretical perspective on how to best teach for engagement and understanding An examination of some of the more contentious debates about pedagogy leads to the advice to seek optimally guided instruction which balances the challenge offered to learners with the level of support provided This provides a framework for discussing a wide range of teaching approaches and techniques that have been recommended to those teaching chemistry across educational levels including both those intended to replace teaching from the front and others that can be built into traditional lecture courses to enhance the learning experience

Kinetic Molecular Theory Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908* **A Model of the Kinetic-Molecular Concept** Slabaugh, 1969-01-01 *A Model of the Kinetic-Molecular Concept* Super Slabaugh, 1969-01-01 **On the Measurement of Kinetic Molecular Energy on an Absolute Scale** George Downing Liveing, 1885

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