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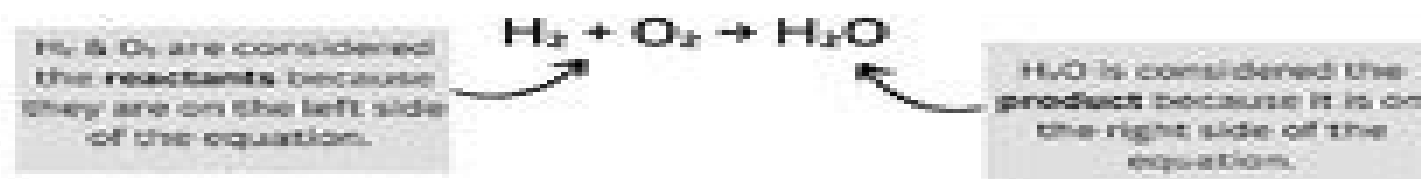
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Balancing Chemical Reactions POGIL

Steps to balancing a chemical reaction:

1. **Writing the Chemical Equation** - Balancing a chemical reaction always starts with writing it out first! A chemical equation shows the **reactants** on the **left side** and the **products** on the **right side** of the reaction arrow.

For example, let's consider the reaction between hydrogen gas (H_2) and oxygen gas (O_2) to produce water (H_2O):

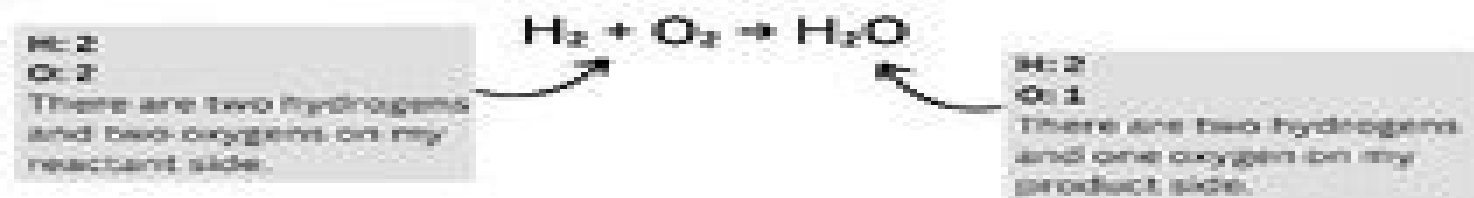


Try identifying the reactants and products of the following chemical equations:



2. **Count the Number of Atoms on Each Side** - Make a list of all atoms on both the reactant side and product side (you will have the same atoms even if they look different). Count individually how many of each atom there are.

For example, let's again consider the reaction between hydrogen gas (H_2) and oxygen gas (O_2) to produce water (H_2O):



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SB Merriam

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'SB Merriam' text.

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