

Polyatomic Ions

Can a group of atoms have a charge?

Why?

Do you know you eat a lot of "-ates"? Next time you look at a food label, read the ingredients and you will likely find a number of ingredients that end with "-ate," such as sodium phosphate or calcium carbonate. Did you ever wonder what the chemical formulas of these ingredients look like? In this activity we will explore polyatomic ions, which are groups of atoms that carry a charge. These ions are found in our food ingredients, natural waterways, and many other chemical compounds you encounter every day.

Model 1 – Types of Ions

Monatomic Ions	Nitride 	Sulfide 	Chloride 
Polyatomic Ions	Nitrate 	Sulfate 	Ammonium 
	Nitrite 	Sulfite 	Hydroxide 

1. Use Model 1 to complete the table below.

Name of Ion	Nitride	Nitrate	Sulfate	Sulfite	Ammonium
Charge on Ion	-3	-1	-2	-2	+1
Type and Number of Atoms	1 nitrogen	1 nitrogen 3 oxygen	1 sulfur 4 oxygen	1 sulfur 3 oxygen	1 nitrogen 4 hydrogen
Chemical Formula	N^{3-}	NO_3^{1-}	SO_4^{2-}	SO_3^{2-}	NH_4^{1+}

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The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills,Amina El-Ashmawy,2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

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Table of Contents Pogil Activities For High School Chemistry Polyatomic Ions Key

1. Understanding the eBook Pogil Activities For High School Chemistry Polyatomic Ions Key
 - The Rise of Digital Reading Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Activities For High School Chemistry Polyatomic Ions Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Personalized Recommendations
 - Pogil Activities For High School Chemistry Polyatomic Ions Key User Reviews and Ratings
 - Pogil Activities For High School Chemistry Polyatomic Ions Key and Bestseller Lists
5. Accessing Pogil Activities For High School Chemistry Polyatomic Ions Key Free and Paid eBooks
 - Pogil Activities For High School Chemistry Polyatomic Ions Key Public Domain eBooks
 - Pogil Activities For High School Chemistry Polyatomic Ions Key eBook Subscription Services
 - Pogil Activities For High School Chemistry Polyatomic Ions Key Budget-Friendly Options

6. Navigating Pogil Activities For High School Chemistry Polyatomic Ions Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pogil Activities For High School Chemistry Polyatomic Ions Key Compatibility with Devices
 - Pogil Activities For High School Chemistry Polyatomic Ions Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Highlighting and Note-Taking Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Interactive Elements Pogil Activities For High School Chemistry Polyatomic Ions Key
8. Staying Engaged with Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil Activities For High School Chemistry Polyatomic Ions Key
9. Balancing eBooks and Physical Books Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Activities For High School Chemistry Polyatomic Ions Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Setting Reading Goals Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Fact-Checking eBook Content of Pogil Activities For High School Chemistry Polyatomic Ions Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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