

O&M Manual

O&M manual is a living document. It should be flexible to accommodate changes, periodically updated based on experience and as organization matures.

O&M manual may include topics such as:

- Installation and configuration operations and associated definitions
- System maintenance, updates and upgrade policies, procedures, and schedules
- Proper and improper handling and maintenance of different types of equipment
- Database scheme, network typology, and flowcharts used to illustrate items such as system design, data communications, program logic, and the relationship between network nodes.
- Instructions for opening/closing and starting/stopping applications, devices, and services under various conditions
- Procedure and sequences describing backup routines, media type, storage locations, and schedules
- Security controls for staff, facilities, infrastructure and emergency response procedures
- Frequently asked questions and troubleshooting techniques for common issues
- Roles, responsibilities and contact information for key personnel and support staff
- Other miscellaneous and/or relevant items

O M Manual Construction Documents

JA Banks



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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8.00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web in this lab we will use experimental evidence and the principles of stoichiometry to determine which ion is formed in a single replacement reaction between iron and copper ii sulfate pre lab assignment in preparation for the lab activity we will investigate another single replacement reaction between copper and silver nitrate

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction

or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalization of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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