

WORD EQUATIONS

Name _____

Write and balance the following chemical equations.

1. Hydrogen plus oxygen yield water.
2. Nitrogen plus hydrogen yield ammonia.
3. Aluminum bromide plus chlorine yield aluminum chloride and bromine.
4. Hydrochloric acid plus sodium hydroxide yield sodium chloride plus water.
5. Iron plus lead (II) sulfate react forming iron (II) sulfate plus lead.
6. Potassium chlorate when heated produces potassium chloride plus oxygen gas.
7. Sulfuric acid decomposes to form sulfur trioxide gas plus water.
8. Sodium oxide combines with water to make sodium hydroxide.
9. Potassium iodide reacts with bromine forming potassium bromide plus iodine.
10. Sodium phosphate reacts with calcium nitrate to produce sodium nitrate plus calcium phosphate.
11. Zinc reacts with iron (III) chloride yielding zinc chloride plus iron precipitate.
12. Ammonium carbonate and magnesium sulfate react to yield ammonium sulfate plus magnesium carbonate.
13. Phosphoric acid plus calcium hydroxide react forming solid calcium phosphate plus water.
14. Aluminum plus oxygen gas form aluminum oxide under certain conditions.
15. Nitrogen gas plus oxygen gas react and form dinitrogen pentoxide.

Physical Science If8767 Word Equations Answers

Sasha Hu



Physical Science If8767 Word Equations Answers:

Physical Science If8767 Word Equations Answers Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has become much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Physical Science If8767 Word Equations Answers**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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