

GRADE 10 PHYSICAL SCIENCES

NOVEMBER EXAMINATION PAPER 2: CHEMISTRY

SECTION A
QUESTION 1

Various options are provided as possible answers to the following questions. Choose the correct answer and write **ONLY** the LETTER (A-D) next to the QUESTION NUMBER in your answer book.

1.1 Which of the following is an example of a mixture?

- A distilled water
- B diamond
- C sodium chloride
- D syngas

1.2 Element with the electron configuration: $1s^2, 2s^2, 2p^4$

- A oxygen
- B carbon
- C helium
- D nitrogen

1.3 The chemical formula of sodium sulfide

- A NaS
- B Na₂S
- C NaS₂
- D Na₂S₂

1.4 Strongest intermolecular force found in water?

- A Ion-dipole
- B Dipole-dipole
- C Van der Waals
- D Hydrogen bond

1.5 The particles of a gas ...

- A are tightly packed.
- B vibrate against each other.
- C move freely and rarely touch.
- D move over one another.

1.6 Element with the highest electronegativity?

- A Hydrogen
- B Helium
- C Oxygen
- D Fluorine

1.7 When an atom of Group 16 ionises to become an ion, the

- A mass number of the atom will increase
- B atomic number of the atom will decrease
- C number of electrons will decrease
- D number of electrons will increase

1.8 An element with the following Aufbau diagram will be classified as a:



- A noble gas
- B halogen
- C alkali earth-metal
- D alkali metal

Physical Science Grade 10 Exemplar 2013 November Paper 2

Barbara Bridgman Perkins



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Table of Contents Physical Science Grade 10 Exemplar 2013 November Paper 2

1. Understanding the eBook Physical Science Grade 10 Exemplar 2013 November Paper 2
 - The Rise of Digital Reading Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade 10 Exemplar 2013 November Paper 2
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Personalized Recommendations
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 User Reviews and Ratings
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 and Bestseller Lists
5. Accessing Physical Science Grade 10 Exemplar 2013 November Paper 2 Free and Paid eBooks
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 Public Domain eBooks
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 eBook Subscription Services
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 Budget-Friendly Options
6. Navigating Physical Science Grade 10 Exemplar 2013 November Paper 2 eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 Compatibility with Devices
 - Physical Science Grade 10 Exemplar 2013 November Paper 2 Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Highlighting and Note-Taking Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Interactive Elements Physical Science Grade 10 Exemplar 2013 November Paper 2
8. Staying Engaged with Physical Science Grade 10 Exemplar 2013 November Paper 2
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade 10 Exemplar 2013 November Paper 2
9. Balancing eBooks and Physical Books Physical Science Grade 10 Exemplar 2013 November Paper 2
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade 10 Exemplar 2013 November Paper 2
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Grade 10 Exemplar 2013 November Paper 2
- Setting Reading Goals Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Grade 10 Exemplar 2013 November Paper 2
- Fact-Checking eBook Content of Physical Science Grade 10 Exemplar 2013 November Paper 2
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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