

- 

- Ⓐ Which one of the following decreases in a group from top to bottom?

51. Metallic nature

- D) Electronegativity

- 
- Diagram illustrating the molecular orbital configuration for the  $N_2^+$  ion. The diagram shows the combination of atomic orbitals (1s, 2s, 2p) from two nitrogen atoms to form molecular orbitals. The resulting molecular orbital configuration is  $1s^2 2s^2 2p^3$ , with a bond order of 2.5.

- 

## SECTION - II

3-2-6-1

10. Pose any two questions to understand the difference between valency and valency electrons.

11. Explain why bonding angle (HOH) in water is  $104^{\circ} 31'$  instead of  $109^{\circ} 28'$ .

# Physical Science 2013 Grade 1paper 2

**Lei Shi**



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