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1.2.4		<p>The molecule is a tetrahedron <input checked="" type="checkbox"/></p> <p>Die molekule is 'n tetraëder <input checked="" type="checkbox"/></p> <p>1 mark/punt</p>	LO2
1.2.5		 <p>Name trigonal pyramid <input checked="" type="checkbox"/></p> <p>Noem trigonale piramidele <input checked="" type="checkbox"/></p> <p>3 marks/punte</p>	LO4
1.3	1.3.1	<p>The electron pair is equally shared between the two Cl atoms <input checked="" type="checkbox"/> and the resulting Cl<sub>2</sub> molecule has no degree of polarity. <input checked="" type="checkbox"/></p> <p>Die elektronpaar word ewe veel gedeel tussen die twee Cl-atome <input checked="" type="checkbox"/> en die resulterende Cl<sub>2</sub>-molekule het geen mate van polariteit nie <input checked="" type="checkbox"/>.</p> <p>2 marks/punte</p>	LO2
	1.3.2	<p>The tendency <input checked="" type="checkbox"/> of an atom to pull the bonded pair <input checked="" type="checkbox"/> of electrons towards itself. <input checked="" type="checkbox"/></p> <p>Die neiging <input checked="" type="checkbox"/> van 'n atoom om die gebinde elektronpare <input checked="" type="checkbox"/> na homself te trek <input checked="" type="checkbox"/>.</p> <p>3 marks/punte</p>	LO2
	1.3.3	<p>Electronegativity of Na = 6    Electronegativity of Cl = 3 Difference is 3 <input checked="" type="checkbox"/></p> <p>With a difference of greater than 2.1 <input checked="" type="checkbox"/> electron transfer will take place and the bond <input checked="" type="checkbox"/> is ionic. <input checked="" type="checkbox"/> (value of 2.1 does not need to be given for this mark)</p> <p>Elektronegatiwiteit van Na = 6    Elektronegatiwiteit van Cl = 3 Verskil is 3 <input checked="" type="checkbox"/></p> <p>Met 'n verskil van meer as 2.1 <input checked="" type="checkbox"/> sal elektroneoordrag plaasvind en die binding is ionies. <input checked="" type="checkbox"/> (waarde van 2.1 hoef nie vir hierdie punt gegee te word nie)</p> <p>4 marks/punte</p>	LO2

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**Mike Jess**



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