

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

Step 8: Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① <div style="text-align: center; font-size: 2em;">B</div> <hr/> <hr/>	P= ② N= _____ E= _____	③ O S O L O G
	M.P. = _____ B.P. = ④ _____	
Properties ⑤ Uses ⑥ Bohr Diagram Lewis Structure <div style="display: flex; align-items: center; justify-content: space-around;">  ⑦ <div style="text-align: center; font-size: 1.5em;">B</div> </div>		

Physical Science If8767 Calculating Current Answers

David Kirk



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web then cascade pressure testing will be required i e complete the pressure test on the higher pressure system prior to system closure welding and then repeat full pressure test procedure for the test of the combined systems at the lowest pressure 2 heat exchangers pressure vessels and fired heaters may be included in a

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