

POGIL: Periodic Table Trends

(Adapted from Rush Henrietta CSD)

Why:

The Periodic Table is one of the greatest inventions in the history of man. It allows scientists to predict physical and chemical properties of the elements. Dimitri Mendeleev (a Russian scientist) and Robert Mosley (a British chemist) put together this table in the late 1800's based on properties of the elements known at the time. The trends in these properties as you go across periods and down groups is the subject of this Chem POGIL.

Periodic Table of the Elements

Legend:

- Alkali metals
- Alkaline earth metals
- Transition metals
- Post-transition metals
- Nonmetals
- Metalloids
- Chalcogens
- Halogens
- Noble gases

Success Criteria:

- Understand the meaning of atomic radius, reactivity, electronegativity, ionization energy
- Recognize trends in atomic radius, reactivity, ionization energy and electronegativity as you go across periods and down groups.

Resources:

Reference Table S, Periodic Table, Unit 3 (Chapter 6) Notes, & Chapter 6 in textbook

Prerequisites:

Answer the following questions using the resources above.

- What happens to the number of valence electrons as you go down a group on the periodic table?
- What happens to the number of valence electrons as you go from left to right across a period?
- What happens to the number of energy levels as you go down a group on the periodic table?
- What happens to the number of energy levels as you go from left to right across a period?
- Draw a Lewis structure for a Cl atom.
- Draw a Bohr diagram for the Cl atom. (Draw above!)
- In this lab we will be talking about atomic radius.
 - Look up the atomic radius of Cl on Table S: _____
 - Why is it possible to show the atomic radius using a Bohr diagram but not by using a Lewis structure?
 - On your Bohr diagram draw an arrow to represent the radius of the Cl atom.
- Define *periodic*:
- Define *trend*:

Periodic Trends Answers

J Spring



Periodic Trends Answers:

Regents Exams and Answers: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day **E3**

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occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs *Foundations of College Chemistry, Alternate* Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis High School Chemistry Unlocked The Princeton Review, 2016-10-18 UNLOCK THE SECRETS OF CHEMISTRY with THE PRINCETON REVIEW High School Chemistry Unlocked focuses on giving you a wide range of key lessons to help increase your understanding of chemistry With this book you ll move from foundational concepts to complicated real world applications building confidence as your skills improve End of chapter drills will help test your comprehension of each facet of chemistry from atoms to alpha radiation Don t feel locked out Everything You Need to Know About Chemistry Complex concepts explained in straightforward ways Walk throughs of sample problems for all topics Clear goals and self assessments to help you pinpoint areas for further review Guided examples of how to solve problems for common subjects Practice Your Way to Excellence 165 hands on practice questions seeded throughout the chapters and online Complete answer explanations to boost understanding Bonus online questions similar to those you ll find on the AP Chemistry Exam and the SAT Chemistry Subject Test High School Chemistry Unlocked covers Building blocks of matter Physical behavior of matter Chemical bonding Chemical reactions Stoichiometry Solutions Acids and bases Equilibrium Organic chemistry Radioactivity and more Interplay of Creativity and Giftedness in Science Melissa K. Demetrikopoulos, John L. Pecore, 2015-12-17 This book explores education for juvenile offenders in relation to Passages Academy which is both similar to and representative of many school programs in juvenile correctional facilities Examining the mission and population of this school contributes to an understanding of the ways in which the teachers think about and ultimately act with respect to their detained juveniles students and particularly illustrates how the tension between punishment and rehabilitation is played out in school policies and design By calling attention to the decisions that surround juvenile detention education the extant research concentrates on three main areas first the social political and pedagogical forces that determine who enters the juvenile justice systems second how these court involved youths are educated while they are in the system and third the practical problems and the social justice issues youths encountered when transitioning back to their community schools I Hope I Don t See You Tomorrow is both heartwarming and heartbreaking its vast empathy for the students that L A Gabay teaches is edifying while its unsparing examination of the forces that push youth into detention is soul shearing Gabay is at once Tocqueville and Kozol he brilliantly guides us through the educational territory that is foreign to most of us even as he paints a searing portrait of teachers who shape lesson plans for students who must

learn under impossible conditions Gabay's haunting and eloquent missive from the front lines of pain and possibility couldn't be more timely as the nation's first black president seeks to lessen the stigma of nonviolent ex-offenders in our society Gabay's book confronts the criminal justice system at its institutional roots in the economic misery and racial strife of schooling that compounds the suffering of poor youth as they are contained by a state that often only pays attention to them when they are in trouble Gabay opens eyes and vexes minds with this stirring and sober account of what it means to teach those whom society has deemed utterly expendable Michael Eric Dyson author of *The Black Presidency Barack Obama and the Politics of Race in America* As a beneficiary of Lee Gabay and his colleague's patience discipline and compassionate teaching at the school this timely book beautifully decrypts the pedagogical framework within the juvenile justice system As America comes to term with its zeal for incarceration policymakers educators government officials parents and advocates should take advantage of this carefully written book and use it as reflection and pause as we prepare our young court-involved students towards adulthood Jim St Germain Advisory counsel on President Obama's Taskforce on Police Community Relations and Mayor Bloomberg's Close to Home initiative *The Good Student* Charles Jay Rop,1996 **Cracking the AP Chemistry Exam, 2014 Edition** Princeton Review (Firm),Paul Foglino,2013-08-06 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests a subject review for all topics and sample questions and answers **Foundations of Inorganic Chemistry** Gary Wulfsberg,2017-11-02 Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence Foundations of Inorganic Chemistry by Gary Wulfsberg is our newest entry into the field of Inorganic Chemistry textbooks designed uniquely for a one semester stand alone course or to be used in a full year inorganic sequence By covering virtually every topic in the test from the 2016 ACS Exams Institute this book will prepare your students for success The new book combines careful pedagogy clear writing beautifully rendered two color art and solved examples with a broad array of original chapter ending exercises It assumes a background in General Chemistry but reviews key concepts and also assumes enrollment in a Foundations of Organic Chemistry course Symmetry and molecular orbital theory are introduced after the student has developed an understanding of fundamental trends in chemical properties and reactions across the periodic table which allows MO theory to be more broadly applied in subsequent chapters Use of this text is expected to increase student enrollment and build students appreciation of the central role of inorganic chemistry in any allied field Key Features Over 900 end of chapter exercises half answered in the back of the book Over 180 worked examples Optional experiments demos Clearly cited connections to other areas in chemistry and chemical sciences Chapter opening biographical vignettes of noted scientists in Inorganic Chemistry Optional General Chemistry review sections Originally rendered two color illustrations throughout *AP Chemistry For Dummies* Peter J. Mikulecky,Michelle Rose Gilman,Kate Brutlag,2008-11-13 A practical and hands on guide for learning the practical

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Chemistry: Molecules, Matter, and Change Media Activities Book Loretta Jones, Carl Hoeger, Peter William Atkins, Regina Schoenfield-Tacher, 2000-01-15 The Media Activity Book MAB for Jones Atkins Chemistry molecules matter and change contains chapters with lists and descriptions of some of the media available as you study the chapter Each activity begins with a specific textbook reference Then you are given a time estimate of how long it will take to use the media An M media icon in the margin of the textbook means that media exists to support that area of text The media is found in three different places on the website and on two CDs **Cambridge International AS and A Level Chemistry Workbook with CD-ROM**

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