

POGIL: Periodic Trends in Atomic Properties

Information

Properties such as the size of an atom (atomic radius), the energy required to remove an electron from an atom (ionization energy), and the energy associated with the addition of an electron to a gaseous atom (electron affinity) can be understood in terms of the electron configuration of the atom and a competition between electron-nucleus attraction and electron-electron repulsion.

Preliminary Activity - Complete the orbital diagrams for the ground state configuration.

	1s	2s	2p		1s	2s	2p
H				C			
He				N			
Li				O			
Be				F			
B				Ne			

Model 1: Electron Ionization & Capture and Electrostatic Interactions

Reaction Equation for Electron Ionization: $X \rightarrow X^+ + e^-$ $\Delta E = 1^{\text{st}}$ ionization energy

Reaction Equation for Electron Capture: $X + e^- \rightarrow X^-$ $\Delta E = \text{electron affinity}$

Dominant inter-particle forces in helium:



Dominant inter-particle forces in lithium:



Periodic Trends And Atomic Properties Pogil Answers

Costel Rizescu, Mihaela Rizescu



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Table of Contents Periodic Trends And Atomic Properties Pogil Answers

1. Understanding the eBook Periodic Trends And Atomic Properties Pogil Answers
 - The Rise of Digital Reading Periodic Trends And Atomic Properties Pogil Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends And Atomic Properties Pogil Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends And Atomic Properties Pogil Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends And Atomic Properties Pogil Answers
 - Personalized Recommendations
 - Periodic Trends And Atomic Properties Pogil Answers User Reviews and Ratings

- Periodic Trends And Atomic Properties Pogil Answers and Bestseller Lists
- 5. Accessing Periodic Trends And Atomic Properties Pogil Answers Free and Paid eBooks
 - Periodic Trends And Atomic Properties Pogil Answers Public Domain eBooks
 - Periodic Trends And Atomic Properties Pogil Answers eBook Subscription Services
 - Periodic Trends And Atomic Properties Pogil Answers Budget-Friendly Options
- 6. Navigating Periodic Trends And Atomic Properties Pogil Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends And Atomic Properties Pogil Answers Compatibility with Devices
 - Periodic Trends And Atomic Properties Pogil Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends And Atomic Properties Pogil Answers
 - Highlighting and Note-Taking Periodic Trends And Atomic Properties Pogil Answers
 - Interactive Elements Periodic Trends And Atomic Properties Pogil Answers
- 8. Staying Engaged with Periodic Trends And Atomic Properties Pogil Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends And Atomic Properties Pogil Answers
- 9. Balancing eBooks and Physical Books Periodic Trends And Atomic Properties Pogil Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends And Atomic Properties Pogil Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends And Atomic Properties Pogil Answers
 - Setting Reading Goals Periodic Trends And Atomic Properties Pogil Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends And Atomic Properties Pogil Answers
 - Fact-Checking eBook Content of Periodic Trends And Atomic Properties Pogil Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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