

Annotated periodic table of the elements

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H 1.01 Hydrogen	Main common oxidation states, electronegativity (above-Pauling)																2 He 4.003 Helium
2 Li 6.94 Lithium	3 Be 9.012 Beryllium																
3 Na 22.99 Sodium	4 Mg 24.31 Magnesium																
4 K 39.10 Potassium	5 Ca 40.08 Calcium																
5 Rb 85.47 Rubidium	6 Sr 87.62 Strontium																
6 Cs 132.91 Cesium	7 Ba 137.33 Barium																
7 Fr 223.028 Francium	8 Ra 226.025 Radium																
Atomic number and symbol above, weight and name below		9 Sc 44.96 Scandium	10 Ti 47.87 Titanium	11 V 50.94 Vanadium	12 Cr 52.00 Chromium	13 Mn 54.94 Manganese	14 Fe 55.85 Iron	15 Co 58.93 Cobalt	16 Ni 58.69 Nickel	17 Cu 63.55 Copper	18 Zn 65.38 Zinc	19 Ga 69.72 Gallium	20 Ge 72.63 Germanium	21 As 74.92 Arsenic	22 Se 78.96 Selenium	23 Br 79.90 Bromine	24 Kr 83.80 Krypton
		25 Y 88.91 Yttrium	26 Zr 91.22 Zirconium	27 Nb 92.91 Niobium	28 Mo 95.94 Molybdenum	29 Tc 98.91 Technetium	30 Ru 101.07 Ruthenium	31 Rh 101.07 Rhodium	32 Pd 106.42 Palladium	33 Ag 107.87 Silver	34 Cd 112.41 Cadmium	35 In 114.82 Indium	36 Sn 118.71 Tin	37 Sb 121.76 Antimony	38 Te 127.60 Tellurium	39 I 126.91 Iodine	40 Xe 131.29 Xenon
		57 to 70		71 La 138.91 Lanthanum	72 Ce 140.12 Cerium	73 Pr 140.91 Praseodymium	74 Nd 144.24 Neodymium	75 Pm 144.91 Promethium	76 Sm 150.36 Samarium	77 Eu 151.96 Europium	78 Gd 157.25 Gadolinium	79 Tb 158.93 Terbium	80 Dy 162.50 Dysprosium	81 Ho 164.93 Holmium	82 Er 167.26 Erbium	83 Tm 168.93 Thulium	84 Yb 173.05 Ytterbium
		89 to 102		89 La 138.91 Lanthanum	90 Ce 140.12 Cerium	91 Pr 140.91 Praseodymium	92 Nd 144.24 Neodymium	93 Pm 144.91 Promethium	94 Sm 150.36 Samarium	95 Eu 151.96 Europium	96 Gd 157.25 Gadolinium	97 Tb 158.93 Terbium	98 Dy 162.50 Dysprosium	99 Ho 164.93 Holmium	100 Er 167.26 Erbium	101 Tm 168.93 Thulium	102 Yb 173.05 Ytterbium
		103 Lu 174.96 Lutetium	104 Hf 178.49 Hafnium	105 Ta 180.95 Tantalum	106 W 183.85 Tungsten	107 Re 186.21 Rhenium	108 Os 190.23 Osmium	109 Ir 192.22 Iridium	110 Pt 195.08 Platinum	111 Au 196.97 Gold	112 Hg 200.59 Mercury	113 Tl 204.38 Thallium	114 Pb 207.2 Lead	115 Bi 208.98 Bismuth	116 Po 209 Polonium	117 At 209 Astatine	118 Rn 222 Radon
		103 Lu 174.96 Lutetium	104 Hf 178.49 Hafnium	105 Ta 180.95 Tantalum	106 W 183.85 Tungsten	107 Re 186.21 Rhenium	108 Os 190.23 Osmium	109 Ir 192.22 Iridium	110 Pt 195.08 Platinum	111 Au 196.97 Gold	112 Hg 200.59 Mercury	113 Tl 204.38 Thallium	114 Pb 207.2 Lead	115 Bi 208.98 Bismuth	116 Po 209 Polonium	117 At 209 Astatine	118 Rn 222 Radon

- Group names*
- 1 Hydrogen & the alkali metals
 - 2 Alkaline earth metals
 - 3 Lanthanoids
 - 4 Actinoids
 - 5 Boron group
 - 6 Chromium group
 - 7 Manganese group
 - 8 Iron group
 - 9 Cobalt group
 - 10 Nickel group
 - 11 Copper group
 - 12 Zinc group
 - 13 Boron group
 - 14 Carbon group
 - 15 Nitrogen group
 - 16 Oxygen group
 - 17 Halogens
 - 18 Helium group

*Groups 3 to 10 are named after their first members i.e. Group 3 is the Scandium Group

CLASS COUNTS/PARTS

Metals	Nonmetals	
Active	Pre-transition	Composite
	Alkali earth Alkali	
Transition	Intermediate	
Fluoride	Metalloid	
None (Fr-Fr, Cs-Ac)		
None gas		

89 La 138.91 Lanthanum	90 Ce 140.12 Cerium	91 Pr 140.91 Praseodymium	92 Nd 144.24 Neodymium	93 Pm 144.91 Promethium	94 Sm 150.36 Samarium	95 Eu 151.96 Europium	96 Gd 157.25 Gadolinium	97 Tb 158.93 Terbium	98 Dy 162.50 Dysprosium	99 Ho 164.93 Holmium	100 Er 167.26 Erbium	101 Tm 168.93 Thulium	102 Yb 173.05 Ytterbium
103 Lu 174.96 Lutetium	104 Hf 178.49 Hafnium	105 Ta 180.95 Tantalum	106 W 183.85 Tungsten	107 Re 186.21 Rhenium	108 Os 190.23 Osmium	109 Ir 192.22 Iridium	110 Pt 195.08 Platinum	111 Au 196.97 Gold	112 Hg 200.59 Mercury	113 Tl 204.38 Thallium	114 Pb 207.2 Lead	115 Bi 208.98 Bismuth	116 Po 209 Polonium

Radioactive
Approximate data for either a metastable or a ground state is given

Active metals

Pre-transition metals
Light, reactive, mostly of low mechanical strength, melting and boiling points. The group 1 members are easily cut with a knife. Only Be, Mg and Al have any structural uses. Most form basic oxides (Be and Al are amphoteric).

Alkali earths
Closely resemble Ca, Sr and Ba, but many are heavier, and they form mostly pale-colored compounds. Similar to one another, esp. La to Lu (aka the lanthanides), and hard to separate for that reason. Abundant but widely dispersed; commercially viable concentrations are rare. Oxides are strongly to moderately basic. Sc, Yb and Lu exhibit weak amphotericism.

Alkalies
Soft, dense and reactive metals. Those from Fr onwards are synthetic. Early alkalies (Cs-Ac) show some similarities to transition metals and have basic, amphoteric, or acidic oxides; late alkalies are more like lanthanides. Many alkalies form coloured compounds.

Transition metals
Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from Rf onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.

Noble metals
Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very weakly electropositive; inertlike oxides.

Frontier metals
Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or lower nearest neighbours than other metals. All used or tried as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterized, to varying degrees, by covalent bonding tendencies, amphotericism, and anionic species such as zincates, stannates, and bismuthates.

Metals

Nonmetals

Metalloids
Metallic looking brittle, fair electrical conductors. All known in amorphous forms. Moderate ionisation energies (750-1,000 kJ/mol), and electronegativity values (1.8-2.3). Mostly behave chemically as non-metals. Can form alloys with metals. Amphoteric or weakly acidic oxides.

Intermediate nonmetals
H and N are colourless gases, C, P, and Se are metalloids looking solids while Si is yellow. The latter four are brittle but all known in malleable, plastic or ductile forms. Moderate to high ionisation energies, low to high electron affinities, moderate to high electronegativity values, relatively poor to moderately strong oxidising agents, tend to form covalent or metalloids to near-metallic compounds with metals. Their oxides are acidic, or neutral (H₂O, CO, NO, N₂O).

* as liquid P, the most stable form in ambient conditions

Group 1
The starts as Sr-Fr, and then bifurcates into a La branch (i.e. La-Ac), and a Lu branch (i.e. Lu-Lr). Trends going down Sr-Fr, and the Lu branch, resemble those of groups 1-2.

* Lu branch, more resemble those of groups 4-10

Composite nonmetals
O, F, and Cl are, respectively, colorless, pale yellow, and pale yellow-green gases. Br is red-brown liquid. I is a metallic looking solid. Chemically tend to have to high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically composite nature of the non-metallic halogens. All five tend to form predominantly ionic compounds with metals. Their oxides are acidic.

Radiative gases
Colourless, colorless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition XePtF₆, was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesised in the ensuing half-century. Compounds of He, Ne, and Ar, in ambient conditions, are not known.

Periodic Table Unscrambled Table Answers

RS Peters



Periodic Table Unscrambled Table Answers:

Microsoft Word Simple Projects Jan Rader, Jan Ray, 2001-04 Projects for language arts social studies science and math Provided templates can be modified to meet specific needs Project samples also provided **Periodic Table** Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket **Unscramble Your Nest Egg** John J. Cunningham, 2013-10-25 Pay off mortgage by age 50 55 deposit former mortgage payments into income earning accounts and result will be a cash nest egg big enough to pay 50 percent of living expenses during 20 years of retirement **Scientifica**, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities *New Modular Science for GCSE*. Ann Fullick, 1998 This homework book provides graded homework questions to help the teacher set homework for a range of abilities It offers 14 sets of homework questions for each of the 12 modules in the NEAB science syllabus **Scientifica Pupil Book 8 (Levels 4-7)** David Sang, Lawrie Ryan, Jane Taylor, Peter Ellis, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities *Cyber Science 6 Tm' 2007 Ed.*, **The Science Teacher**, 2008 *Pathways in Science: Chemistry (Pt. 1-3)* Joseph M. Oxenhorn, 1970 Story of the World Activity Book 4 Modern Age Susan Wise Bauer, 2005-11-29 Presents a history of the ancient world from 6000 B C to 400 A D **Amazing People: Scientists and Mathematicians** Elise Craver, Malaski, Malay, Hailey Scragg, 2023-01-23 Inspirational 1st 2nd and 3rd grade workbooks for kids ages 7 Motivate and inspire children with the Amazing People Scientists and Mathematicians Activity Book 1st 2nd and 3rd grade books are a great way for children to learn about inspirational people who have made important contributions to science and math through a variety of activities that are both fun AND educational Why You ll Love This 1st Grade Third Grade Workbook Engaging and educational activities and wall d cor The activity book features biographies activities and flash cards covering 53 inspirational mathematicians and scientists The activity book also includes 1 inspirational poster great for both homeschool and classroom curriculum and classroom d cor Tracking progress along the way Test your child s knowledge with the flash cards that cover scientists and mathematicians learned throughout the book Use the stickers to reward students on a job well done after completing activities Practically sized for every activity The 256

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Jumpstarters for Properties of Matter, Grades 4 - 12 Eric T. Olson,2008-12-19 Engage students in describing and comparing general properties of objects like size shape and temperature Then move on to specific types of matter that have characteristic properties such as density and the melting point And lastly have students focus on the notions of elements and compounds Each reproducible page includes five exercises that can be used as whole worksheets for homework assignments and extra practice or cut apart for daily warm up activities Supports NSE standards **Verbal Reactions - Word**

Scrambles With a Chemical Flavor (Easy) Chris McMullen,Carolyn Kivett,2011-12-09 VERBAL ReAcTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve VERBAL ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions EXAMPLE Here is a sample Es 2 S P Si 2 O N _ _ _ _ _ _ _ _ This VERBAL ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that s why there are 8 blanks in the puzzle ANSWER For this puzzle the answer is P O S S Es Si O N possession These VERBAL ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VERBAL ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Easy volume consists of words with 4 to 5 symbols which involves familiarity with common 4 to 10 letter words Other Medium and Hard volumes consist of longer words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U _ _ _ _ _ 2 2 C N 2 I P _ _ _ _ _ 3 Ti C Cr P Y _ _ _ _ _ 4 2 C U 2 S Es _ _ _ _ _ You can find the answers below Note that this easy volume consists of chemical words with 4 to 5 symbols We recommend starting with our easy puzzles before tackling the medium or hard puzzles available in separate volumes ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS KIRIBATI NARAYAN CHANGDER,2023-01-13 IF YOU ARE LOOKING FOR A FREE PDF PRACTICE SET OF THIS BOOK FOR YOUR STUDY PURPOSES FEEL FREE TO CONTACT ME cbsenet4u gmail com I WILL SEND YOU

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Verbal Reactions - Word Scrambles With a Chemical Flavor (Hard) Chris

Mcmullen,Carolyn Kivett,2011-06-18 VErBAL ReAcTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve VErBAL ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions Here is a sample Es 2 S P Si 2 O N _ _ _ _ _ This VErBAL ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that s why there are 8 blanks in the puzzle For this puzzle the answer is P O S S Es Si O N possession These VErBAL ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VErBAL ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Hard volume consists of words with 7 to 8 symbols which involves familiarity with common 8 to 14 letter words Other Medium and Easy volumes consist of shorter words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U _ _ _ _ _ 2 2 C N 2 I P _ _ _ _ _ 3 Ti C Cr P Y _ _ _ _ _ 4 2 C U 2 S Es _ _ _ _ _ You can find the answers at the end of this paragraph Note that this hard volume consists of chemical words with 7 to 8 symbols which are longer than the examples shown here We recommend starting with our easy or medium puzzles before tackling these hard puzzles available in separate volumes ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS

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Directory of publishers **Science Reporter** ,2001 **FCC Record** United States. Federal Communications
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