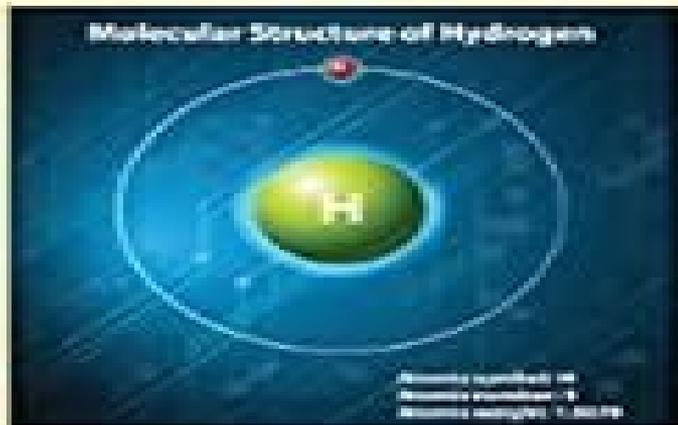


PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Basic Information

Name: Hydrogen
Symbol: H
Atomic Number: 1
Atomic Mass: 1.00794 amu
Melting Point: -259.14 °C (-314.654695 K, -434.41203 °F)
Boiling Point: -252.87 °C (-201.040495 K, -423.146 °F)
Number of Protons/Electrons: 1
Number of Neutrons: 0
Classification: Non-metal
Crystal Structure: Hexagonal
Density @ 293 K: 0.08989 g/cm³
Color: colorless

Atomic Structure

Number of Energy Levels: 1
First Energy Level: 1

Isotopes

| Isotope | Half Life |
|---------|------------|
| H-1 | Stable |
| H-2 | Stable |
| H-3 | 12.5 years |

Facts

Name Origin: Greek words *hydro* (water) and *genes* (generated)
Uses: Balloons, metal refining
Obtained From: natural, H₂ gas water

Molecular Structure of Carbon



Name: Carbon

Symbol: C
Atomic Number: 6
Atomic Mass: 12.0107 amu
Melting Point: 3500.0 °C (3273.25 K, 6332.0 °F)
Boiling Point: 4827.0 °C (5100.25 K, 8700.0 °F)
Number of Protons/Electrons: 6
Number of Neutrons: 6
Classification: Non-metal
Crystal Structure: Hexagonal
Density @ 293 K: 2.02 g/cm³
Color: May be black

Atomic Structure

Number of Energy Levels: 2
First Energy Level: 2 **Second Energy Level:** 4

Isotopes

| Isotope | Half Life |
|---------|--------------|
| C-11 | 20.3 minutes |
| C-12 | Stable |
| C-13 | Stable |
| C-14 | 5730.0 years |
| C-15 | 2.3 seconds |

Facts

Uses: steel, diamonds
Obtained From: burning with insufficient oxygen



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