## CHEMISTRY

## Periodic Properties and Variation of Properties

**DHA-02** 

- Q1 Which of the following elements has 2 shells, and both are completely filled?
  - (A) Helium

(B) Neon

(C) Calcium

- (D) Boron
- Q2 Why group one elements are called alkali metals?
  - (A) When combined with water, their oxides produce acidic solutions.
  - (B) When their oxides and hydroxides are treated with water, they produce alkaline solutions.
  - (C) When their peroxides are treated with water. they produce alkaline solutions.
  - (D) When treated with water, their hydroxides produce acidic solutions.
- Q3 Which of the following gives the correct order of atomic radii of O. F and N?
  - (A) O. F. N.

(B) N. F. O.

(C) O, N, F

- (D) F.O.N.
- Q4 Where would you locate the element with electronic configuration 2.8 in the Modern Periodic Table?
  - (A) Group 8

(B) Group 2

(C) Group 18

- (D) Group 10
- Q5 The order of increasing atomic size is:
  - (A) I < Br < Cl < F
  - (B) I > Br > Cl > F
  - (C) Br > I > Cl > F

- (D) Cl > Br > F > I
- Q6 Newlands arranged the elements known at that time in an increasing order of their

masses.

- (A) electrons
- (B) protons
- (C) atomic
- (D) ions
- Q7 As per Mendeleev periodic table properties of elements are periodic function of their
  - (A) Atomic number
  - (B) Atomic mass
  - (C) Molecular mass
  - (D) all of the above
- Q8 Demerits of Mendeleev periodic table are
  - (A) position of Hydrogen
  - (B) Position of Isotopes
  - (C) Position of Co and Nickel
  - (D) all of the above
- Q9 Who made modern periodic table
  - (A) Mendeleev
- (B) John Dalton
- (C) John Newland
- (D) Henry Moseley
- Q10 The entire modern periodic table can be divided into how many blocks
  - (A) 2 blocks

(B) 3 blocks

(C) 4 blocks

(D) 5 blocks





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# **Periodic Variation Of Properties Answer Key**

**L** Towne

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