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4.14.

$$\left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g/g}$$

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

4.15. (a) 100 g water contains 100 g of water.

- (b) The water contains 100 g of water. The water contains 100 g of water.  
 (c) The water contains 100 g of water. The water contains 100 g of water.

4.16. (a) 100 g of water contains 100 g of water. The water contains 100 g of water.

$$\text{mass of H}_2\text{O} = 100 \text{ g} \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

4.17. (a) 100 g of water contains 100 g of water. The water contains 100 g of water.

$$\text{mass of H}_2\text{O} = 100 \text{ g} \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

$$\text{mass of H}_2\text{O} = 100 \text{ g} \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

4.18. (a) 100 g of water contains 100 g of water. The water contains 100 g of water.

$$\text{mass of H}_2\text{O} = 100 \text{ g} \left( \frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

# Physical Chemistry Ball Solution Manual

**Peter William Atkins**



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