

Periodic Trends Worksheet

Use the periodic table and your knowledge of periodic trends to answer the following questions.

1. Which atom in each pair has the larger atomic radius?

a) Li or **K** b) **Ca** or Ni c) **Ga** or B d) O or **C** e) Cl or **Br**
f) Be or **Ba** g) **Si** or S h) Fe or **Au**

2. What is the periodic trend for atomic size from top to bottom in a group? from left to right in a period?

Top to bottom: increases due to adding energy levels. Left to Right: decreases

3. Why do atoms get smaller as you move left to right in a period?

due to increased nuclear charge and no added shielding.

4. Which element in each pair has a larger ionization energy?

a) Na or **O** b) **Be** or Ba c) **Ar** or F d) **Cu** or Ra e) I or **Ne**
f) K or **V** g) **Ca** or Fr h) W or **Se**

5. Explain the relationship between the relative size of an ion to its neutral atom and the charge on the ions.

Positive ions are smaller than the neutral atom it comes from, negative ions are larger than the neutral atom it comes from

6. Which particle has the larger radius in each atom/ion pair?

a) Na, Na⁺ b) S, S²⁻ c) I, I⁻ d) Al, Al³⁺

7. What is ionization energy? What is first ionization energy?

The amount of energy needed to remove an electron from an atom. First ionization energy is the energy required to remove a single valence electron from an atom.

8. What is the periodic trend for first ionization energy?

It increases going from left to right due to increased nuclear charge, and decreases going down a group due to increased distance from the nucleus and increased shielding.

9. Arrange the following groups of elements in order of increasing ionization energy.

a) Be, Mg, Sr b) Bi, Cs, Ba c) Na, Al, S
high, middle, low high, low, middle low, middle, high

10. Which element in each pair has a higher electronegativity value?

a) Cl, **F** b) C, **N** c) **Mg**, Ne d) **As**, Ca

11. What is the periodic trend for electronegativity? Increases going left to right, decreases going down.

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Yan Bai



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web nfpa 17a standard for wet chemical extinguishing systems 2009 edition international building code 2015 code of federal regulations title 46 shipping pt 166 199 revised as of october 1 2009

nfpa 17a 2009 secure4 khronos - Jan 29 2022

nfpa standartları finder elektronik güvenilir teknoloji - Apr 12 2023

web nfpa 17a ıslak kimyasal söndürme sistemleri standardı nfpa 18 ıslatma maddelerinde standart nfpa 18a yangın kontrolü ve buhar azaltma için su katkıları standardı

nfpa fire 17a techstreet - Dec 08 2022

web nfpa 17a is a must have for anyone who purchases designs installs tests inspects approves lists operates or maintains pre engineered wet chemical fire extinguishing equipment major revisions to the 2021 edition include addition of a chapter addressing wet chemical extinguishing systems for mobile equipment

nfpa 17a standard for wet chemical extinguishing systems - Aug 16 2023

web standard for wet chemical extinguishing systems this standard includes the minimum requirements to ensure that pre engineered wet chemical fire extinguishing systems will function as intended throughout their life to protect life and property from fire

nfpa 17a 2009 madcad com - Feb 27 2022

web jun 12 2023 paint booth nfpa 17 2009 kitchen hoods nfpa 17a 2009 construction codes per 2009 nfpa 17a section 5 2 1

8 and ofc 904 3 4 an audible visual alarm shall be provided to show that the system has operated and that the system is in need of a recharge notes for hood extinguishing systems 2009 nfpa 17a 2008

ul 300 kitchen fire suppression system requirements nfpa 17a - Jul 03 2022

web jan 22 2018 nfpa 17a requires a full inspection of your system at least twice a year semiannually and after any activation like your monthly inspections these inspections must follow the manufacturer s guidelines