

Annotated periodic table of the elements

1		2		Group name*																13		14		15		16		17		18	
1 H 1.01 Hydrogen				1 Hydrogen & the alkali metals																14 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
3 Li 6.94 Lithium		4 Be 9.01 Beryllium		2 Alkaline earth metals																15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
11 Na 22.99 Sodium		12 Mg 24.31 Magnesium		11 Coinage metals																15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
19 K 39.10 Potassium		20 Ca 40.08 Calcium		12 Volatile metals																15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
37 Rb 85.47 Rubidium		38 Sr 87.62 Strontium		13 Iodogens																15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
55 Cs 132.91 Cesium		56 Ba 137.33 Barium		*Groups 1 to 10 are named after their first members i.e. Group 1 is the Sodium group																15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
87 Fr 223.02 Francium		88 Ra 226.02 Radium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
101 Md 258.10 Mendelevium		102 No 259.10 Nobelium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
103 Lr 260.10 Lawrencium		104 Rf 261.10 Rutherfordium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
105 Db 262.10 Dubnium		106 Sg 263.10 Seaborgium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
107 Bh 264.10 Bohrium		108 Hs 265.10 Hassium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
109 Mt 266.10 Meitnerium		110 Ds 267.10 Darmstadtium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
111 Rg 268.10 Roentgenium		112 Cn 269.10 Copernicium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
113 Nh 270.10 Nihonium		114 Fl 271.10 Flerovium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
115 Mc 272.10 Moscovium		116 Lv 273.10 Livermorium																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	
117 Ts 274.10 Tennessine		118 Og 276.10 Oganesson																		15 Boron		14 Carbon		15 Nitrogen		16 Oxygen		17 Fluorine		18 Neon	

CLASS COUNTERPARTS		
Metals		Nonmetals
Active	Pre-transition Alkali earth Alkaline	Cometoid
Transition		Intermediate
Frontier		Metalloid
Node (Ru-Pt-Co-Au)		Node gas

57 La 138.91 Lanthanum	58 Ce 140.12 Cerium	59 Pr 140.91 Praseodymium	60 Nd 144.24 Neodymium	61 Pm 144.91 Promethium	62 Sm 150.36 Samarium	63 Eu 151.96 Europium	64 Gd 157.25 Gadolinium	65 Tb 158.93 Terbium	66 Dy 162.50 Dysprosium	67 Ho 164.93 Holmium	68 Er 167.26 Erbium	69 Tm 168.93 Thulium	70 Yb 173.05 Ytterbium
89 Ac 227.03 Actinium	90 Th 232.04 Thorium	91 Pa 231.04 Protactinium	92 U 238.03 Uranium	93 Np 237.05 Neptunium	94 Pu 244.06 Plutonium	95 Am 243.06 Americium	96 Cm 247.07 Curium	97 Bk 247.07 Berkelium	98 Cf 251.08 Californium	99 Es 252.08 Einsteinium	100 Fm 257.10 Fermium	101 Md 258.10 Mendelevium	102 No 259.10 Nobelium

Active metals

Pre-transition
Light, reactive, mostly of low mechanical strength, melting and boiling points. The group 1 members are easily cut with a knife. Only Be, Mg and Al have any structural uses. Most form basic oxides (Be and Al are amphoteric).

Rare earths
Closely resemble Ca, Sr and Ba, but many are heavier and they form mostly pale coloured compounds. Similar to one another, esp. La to Lu (aka the lanthanides), and hard to separate for that reason. Abundant but widely dispersed; commercially viable concentrations are rare. Oxides are strongly to moderately basic. Sc, Yb and Lu exhibit weak amphotericism.

Actinides
Soft, dense and reactive metals. Those from Ra onwards are synthetic. Early actinides (Th-Am) show some similarities to transition metals and have basic, amphoteric, or acidic oxides; late actinides are more like lanthanides. Many actinides form coloured compounds.

Transition metals
Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from Rf onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.

Noble metals
Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very easily electroplated; insoluble oxides.

Frontier metals
Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or lower nearest neighbours than other metals. All used or tried as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterised, to varying degrees, by covalent bonding tendencies, amphotericism, and anionic species such as zincates, stannates, and bismuthates.

Metalloids
Metallic looking brittle, fair electrical conductors. All known in semiconducting forms. Moderate ionisation energies (750–1,000 kJ/mol), and electronegativity values (1.8–2.3). Mostly behave chemically as non-metals. Can form alloys with metals. Amphoteric or weakly acidic oxides.

Intermediate nonmetals
H and N are colourless gases; C, P, and Se are metalloids looking solids while S is yellow. The latter four are brittle but all known in malleable, plastic or ductile forms. Moderate to high ionisation energies, low to high electron affinities, moderate to high electronegativity values, relatively poor to moderately strong oxidising agents, tend to form covalent or metallic to near-metallic compounds with metals. Their oxides are acidic, or neutral (H₂O, CO, NO, N₂O).

Group 1
This group is Sc-Y, and the alkaline earths are La-Be, and 4 to 10 (not 11–12). Trends going from Sc-Y and Be-10 (not 11–12).

Group 2
This group is Sc-Y, and the alkaline earths are La-Be, and 4 to 10 (not 11–12). Trends going from Sc-Y and Be-10 (not 11–12).

Cometoid nonmetals
O, F, and Cl are, respectively, colourless, pale yellow, and pale yellow-green gases. Br is a red-brown liquid. I is a metallic looking solid. Chemically tend to have to high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically cometoid nature of the non-metallic halogens. All five tend to form predominantly ionic compounds with metals. Their oxides are acidic.

Noble gases
Colourless, odourless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition XePF₆, was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesised in the ensuing half-century. Compounds of He, Ne, and Ar, in ambient conditions, are not known.

Periodic Table Guide Answer Sheet

Anthony S. Fauci



Periodic Table Guide Answer Sheet:

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In a digitally-driven world wherever screens reign great and quick interaction drowns out the subtleties of language, the profound secrets and mental subtleties concealed within words frequently get unheard. However, nestled within the pages of **Periodic Table Guide Answer Sheet** a charming fictional value pulsing with fresh thoughts, lies an exceptional quest waiting to be undertaken. Penned by an experienced wordsmith, that marvelous opus invites viewers on an introspective trip, gently unraveling the veiled truths and profound affect resonating within the very cloth of each and every word. Within the mental depths of this poignant evaluation, we can embark upon a honest exploration of the book is key subjects, dissect its charming publishing model, and yield to the powerful resonance it evokes strong within the recesses of readers hearts.

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email to request for an electric fan at the office sample email - Jun 17 2023

web jun 10 2022 i am writing you this letter to inform you about the need for new ceiling fans at our office the old fans are very old and are making weird noises also we need more

sample request letter for a electric fan book - Jun 05 2022

web sep 15 2023 sample request letter for a electric fan steely dan fan s bluebook dandom com s guestbook usb fan with led clock wholesale with led suppliers