

Periodic Table Questions and Answers

How many elements are there?

118

What is the periodic table?

A chart that organizes elements by atomic number and shared properties

Who invented the periodic table?

Dmitri Mendeleev

A chart that organizes elements by atomic number and shared properties

| | | | | | | | | | | | | | | | | | | | | | | | |
|-----------------------|------------------------|------------------------|--------------------------|--------------------------|------------------------|------------------------|-----------------------|-----------------------|------------------------|-----------------------|-------------------------|-------------------------|-----------------------|--------------------------|-----------------------|----------------------|---------------------|---------------------|--------------------|----------------------|-------------------|----------------------|-------------------|
| 1 H Hydrogen | 2 He Helium | | | | | | | | | | | | | | | | | 11 B Boron | 12 C Carbon | 13 N Nitrogen | 14 O Oxygen | 15 F Fluorine | 16 Ne Neon |
| 3 Li Lithium | 4 Be Beryllium | | | | | | | | | 5 Na Sodium | 6 Mg Magnesium | | | | | | | 7 Al Aluminum | 8 Si Silicon | 9 P Phosphorus | 10 S Sulfur | 17 Cl Chlorine | 18 Ar Argon |
| 19 K Potassium | 20 Ca Calcium | 21 Sc Scandium | 22 Ti Titanium | 23 V Vanadium | 24 Cr Chromium | 25 Mn Manganese | 26 Fe Iron | 27 Co Cobalt | 28 Ni Nickel | 29 Cu Copper | 30 Zn Zinc | 31 Ga Gallium | 32 Ge Germanium | 33 As Arsenic | 34 Se Selenium | 35 Br Bromine | 36 Kr Krypton | | | | | | |
| 37 Rb Rubidium | 38 Sr Strontium | 39 Y Yttrium | 40 Zr Zirconium | 41 Nb Niobium | 42 Mo Molybdenum | 43 Tc Technetium | 44 Ru Ruthenium | 45 Rh Rhodium | 46 Pd Palladium | 47 Ag Silver | 48 Cd Cadmium | 49 In Indium | 50 Sn Tin | 51 Sb Antimony | 52 Te Tellurium | 53 I Iodine | 54 Xe Xenon | | | | | | |
| 55 Cs Cesium | 56 Ba Barium | 57 La Lanthanum | 58 Ce Cerium | 59 Pr Praseodymium | 60 Nd Neodymium | 61 Pm Promethium | 62 Sm Samarium | 63 Eu Europium | 64 Gd Gadolinium | 65 Tb Terbium | 66 Dy Dysprosium | 67 Ho Holmium | 68 Er Erbium | 69 Tm Thulium | 70 Y Yttrium | 71 Lu Lutetium | | | | | | | |
| 87 Fr Francium | 88 Ra Radium | 89 Ac Actinium | 90 Th Thorium | 91 Pa Protactinium | 92 U Uranium | 93 Np Neptunium | 94 Pu Plutonium | 95 Am Americium | 96 Cm Curium | 97 Bk Berkelium | 98 Cf Californium | 99 Es Einsteinium | 100 Fm Fermium | 101 Md Mendelevium | 102 No Nobelium | | | | | | | | |
| 103 Nh Nihonium | 104 Fl Flerovium | 105 Mc Moscovium | 106 Lv Livermorium | 107 Ts Tennessine | 108 Og Oganesson | | | | | | | | | | | | | | | | | | |

periodic
nds?
r, ionization

Why a
They have

What are the periodic table trends?

electronegativity, ionization energy, atomic radius, electron affinity

Why are noble gas so inert?

They have stable electron configurations.

Periodic Table Test Questions And Answers

R Sanford



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