

Introduction

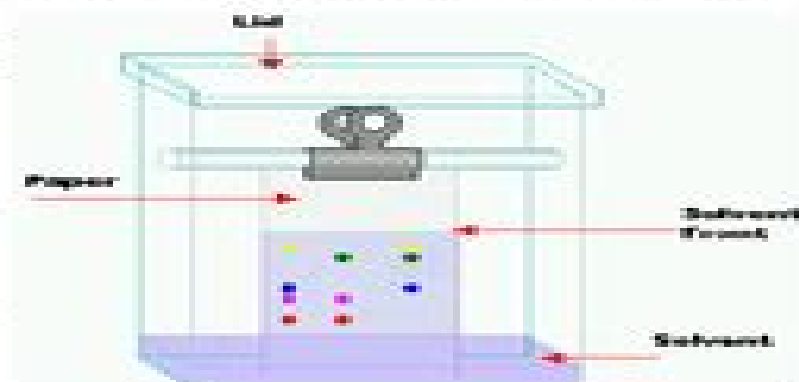
I work for a large chemical company, ChemicaEquip and it has been my job as a technical assistant to carry out different chromatographic techniques to separate and identify components in a mixture and identify different substances. In this report, I will describe and evaluate how I can separate different substances with chromatography. Chromatography is a practical technique used to separate and identify the components in a mixture. Chromatography involves a mixture being dissolved in a mobile phase (which could be a liquid or a gas), that is then passed through an immobile stationary phase (which is usually a solid). Chromatography can be used to separate and identify the components in a mixture. Types of chromatography include:

- Thin-Layer chromatography (TLC) is a technique in which compounds are separated on a thin layer of **adsorbent** material, typically a coating of silica gel on a glass plate or plastic sheet.
- Paper chromatography is an analytical method used to separate coloured chemicals or substances.

There are many advantages as well as disadvantages for paper chromatography and TLC.

Paper chromatography

Paper chromatography is basic, and the stationary phase is usually on paper whilst the mobile phase is the solvent usually water or a solution. The paper has a thin coating of water molecules which may affect the results. The advantages of using paper chromatography are being able to separate water soluble compounds. For example, inks detected by the naked eye. Whereas the disadvantages are that it is limited in terms of mobile phases as it is restricted to only water-soluble compounds. It can also be prone to smearing affecting the validity and reliability of the experiment. Another advantage of paper chromatography is that it requires very less quantitative material and is cheaper compared to other methods such as TLC. Another disadvantage is that there is a large quantity of sample



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