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Roman Wölfel

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web m x c x Δt equation 1 where m is the total mass of the solution solute plus solvent c is the specific heat of the solution and Δt is the observed temperature change the specific heat of the solution is generally assumed to be the same as that of guided inquiry designing the ideal hand warmer - Dec 27 2022

web the heat of reaction Δ hsoln is written after the products in units of kj mol rxn for example sodium hydroxide dissolves exothermically releasing 44 2 kilojoules per mole dissolved this process is represented as naoh s na aq oh aq Δ hsoln 44 2 kj mol rxn write equations to similarly represent the dissolving process for

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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