

Quiz Standard 1 Answer Key

Teacher Name:

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1)



Identify the number of neutrons in one atom of sulfur with an isotopic mass number of 34 amu.

- A) 16
- B) 18
- C) 32
- D) 34
- E) None of the above

Explanation:

One atom of sulfur with an isotopic mass number of 34 has **18** neutrons. 34 (total) - 16 (protons) = 18 (neutrons).

2) Which is true of ionic compounds?

- A) They are good conductors as solids.
- B) They are made by sharing electrons.
- C) They are made of metals and nonmetals.
- D) They are electrically neutral when dissolved in a solution.
- E) None of the above

Explanation:

They are made of metals and nonmetals is true of ionic compounds. They usually form between elements on opposite sides of the periodic table. Ionic compounds are only good conductors when molten or dissolved. When dissolved, they form ions, charged particles. Ionic compounds transfer electrons.

3) Which is true of all covalent compounds?

- A) They share electrons in bonds.
- B) They are electrically charged.
- C) They are good conductors in water.
- D) They are usually made of a metal and a nonmetal.
- E) None of the above

Radioactive Decay 21 1 Answer Key

**John A. Olmsted, Gregory M.
Williams, Robert Charles Burk**



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