

# Page 635 or Appendix D (Page 1044)

**TABLE 16.3 Acid-Dissociation Constants of Some Common Polyprotic Acids**

| Name       | Formula                                    | $K_{a1}$             | $K_{a2}$              | $K_{a3}$              |
|------------|--------------------------------------------|----------------------|-----------------------|-----------------------|
| Ascorbic   | $\text{H}_2\text{C}_6\text{H}_6\text{O}_6$ | $8.0 \times 10^{-5}$ | $1.6 \times 10^{-12}$ |                       |
| Carbonic   | $\text{H}_2\text{CO}_3$                    | $4.3 \times 10^{-7}$ | $5.6 \times 10^{-11}$ |                       |
| Citric     | $\text{H}_3\text{C}_6\text{H}_5\text{O}_7$ | $7.4 \times 10^{-4}$ | $1.7 \times 10^{-5}$  | $4.0 \times 10^{-7}$  |
| Oxalic     | $\text{H}_2\text{C}_2\text{O}_4$           | $5.9 \times 10^{-2}$ | $6.4 \times 10^{-5}$  |                       |
| Phosphoric | $\text{H}_3\text{PO}_4$                    | $7.5 \times 10^{-3}$ | $6.2 \times 10^{-8}$  | $4.2 \times 10^{-13}$ |
| Sulfurous  | $\text{H}_2\text{SO}_3$                    | $1.7 \times 10^{-2}$ | $6.4 \times 10^{-8}$  |                       |
| Sulfuric   | $\text{H}_2\text{SO}_4$                    | Large                | $1.2 \times 10^{-2}$  |                       |
| Tartaric   | $\text{H}_2\text{C}_4\text{H}_4\text{O}_6$ | $1.0 \times 10^{-3}$ | $4.6 \times 10^{-5}$  |                       |

## R59 Appendix D Chemistry Answers

**Peipei Pang**



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