

PRE-AP CHEMISTRY: Semester 2 Final Review

Final: June 8 (period 7); June 9 (period 6)

Review days after school: Monday June 1, Wednesday June 3, Thursday June 4

*You have come a long way this year! You should celebrate your successes and continue to work hard until the end. Let me know how I can help you!

**You can also use a periodic table and your reference sheets from the semester.

***Your binder contains your work from the whole semester (maybe even the whole year!) You can study from your semester's old worksheets, practice problems within the chapters, labs, quizzes, or tests, or look in my folders for extra worksheets/practice.

I. SEMESTER 1:

- a. Significant figures: rules for measuring, rounding, etc.
 - i. Why do we need them?
 - ii. How can you tell how many sig figs are in a measurement?
- b. Scientific notation, converting metric units
 - i. $5.64 \times 10^3 =$ _____
 - ii. $0.000374 =$ _____
 - iii. $5000 \text{ grams} =$ _____ kg
 - iv. $0.345 \text{ L} =$ _____ mL
- c. Physical/chemical changes (in Ch. 9 again)
- d. Structure of an atom: numbers of protons, neutrons, electrons
 - i. What are isotopes?
 - ii. How do you identify the number of protons, neutrons, and electrons in an atom?
- e. Electron configurations
 - i. Use the Aufbau principle to write e^- configs for any element on P-table
 - ii. Know electron configurations for **ions**; what has changed? Why do atoms become ions?
- f. Arrangement of periodic table
 - i. Names of specific groups on table: halogens, alkali metals, noble gases, metals, nonmetals

II. UNIT 5 IONIC BONDING (Chapter 4/12)

- i. How do positive ions form? Negative ions?
- ii. What is the octet rule? How does ionic bonding happen?
- iii. Naming and writing formulas, including those with transition metals
 1. $\text{MgCl}_2 =$ _____
 2. $\text{KF} =$ _____
 3. Iron (III) oxide = _____
- iv. Common polyatomic ions
 1. $\text{Na}_2\text{CO}_3 =$ _____
 2. Aluminum hydroxide = _____

III. UNIT 5: COVALENT BONDING

- a. What is the octet rule? How does covalent bonding happen?
- b. Single, double, triple bonds + valence electrons
- c. Naming and writing formulas (system of prefixes)
- d. VSEPR: predicting shapes of molecules
- e. Polar/nonpolar: what is it, how do you know?
 - i. Draw Lewis structures for the following molecules and identify the shape and polarity:
 1. H_2O
 2. NH_3
 3. CH_4
 4. SF_6
 5. SO_4^{2-}

IV. UNIT 6/7: CHEMICAL REACTIONS

- a. What is a chemical reaction?

Pre Ap English 8 Semester 1 Exam Review

David Baud



Pre Ap English 8 Semester 1 Exam Review:

Resources in Education ,1995 Atlanta Magazine ,2005-01 Atlanta magazine s editorial mission is to engage our community through provocative writing authoritative reporting and superlative design that illuminate the people the issues the trends and the events that define our city The magazine informs challenges and entertains our readers each month while helping them make intelligent choices not only about what they do and where they go but what they think about matters of importance to the community and the region Atlanta magazine s editorial mission is to engage our community through provocative writing authoritative reporting and superlative design that illuminate the people the issues the trends and the events that define our city The magazine informs challenges and entertains our readers each month while helping them make intelligent choices not only about what they do and where they go but what they think about matters of importance to the community and the region **The American School Board Journal** William George Bruce,William Conrad Bruce,1923

The College Board College Handbook College Entrance Examination Board,2007-06 Presents information on enrollment fields of study admission requirements expenses and student activities at two and four year colleges

Cumulated Index Medicus ,1993 2005 College Handbook: More Than 3,600 4-year and 2-year Colleges CollegeBoard,2004 Presents information on enrollment fields of study admission requirements expenses and student activities at two and four year colleges **Resources in Education** ,1990 **Profiles of American Colleges with CD-ROM** Barron's Educational Series,2006-07-01 Up to date information on enrollments tuition and fees academic programs campus environment available financial aid and much more combine to make 27th edition of Profiles of American Colleges America s most authoritative source for information on colleges and universities College bound students parents and high school guidance counselors will find information on more than 1 650 accredited four year colleges across the United States A CD ROM enclosed with each copy of this comprehensive directory presents an interactive format and lets students find individual schools by entering specific criteria In addition to the above cited information each college profile gives details on admission requirements library and computer facilities athletic facilities extracurricular activities e mail addresses fax numbers web sites and more Schools are rated according to Barron s famous competitiveness scale from Noncompetitive to Most Competitive The book s extra section of tinted pages presents a complete quick reference Index of College Majors listing all available major study programs at each school Also profiled are many of the best known colleges in Canada and several other countries The New York Times Index ,1997 **Barron's Profiles of American Colleges** ,2004

Catalog Florida International University,1990 **American Universities and Colleges** ,1964 The College Board College Handbook College Board,2008 This comprehensive guide contains objective information on every accredited college in the U S 2 150 four year colleges and universities and 1 650 two year and community colleges A planning calendar and worksheets help students organize their applications **The Right College, 1991** College Research Group of Concord

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