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Thomas Griffiths

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web answer three questions please insert any standard constants here please note this examination paper is printed on both sides turn over table 1

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web polyatomic ions are charged groups of atoms an example is ammonium ion nh4 it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is nh4cl and ammonium sulfide is nh4 2s

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ion names and ion formulas of the most common polyatomic ions for example no 3 no 3 is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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