

Endothermic and Exothermic reaction Worksheet

Name _____ date _____ period _____

1 Exothermic and endothermic reactions

Decide whether each of these reactions is exothermic or endothermic:

- a) When two chemicals mix their temperature rises: ____exo____
- b) A solid burns brightly and releases heat, light and sound: ____exo____
- c) When two chemicals are mixed their temperature drops: ____endo____
- d) Two chemicals will only react if you heat them continually: ____endo____
- e) Plants take in light energy for photosynthesis: ____endo____

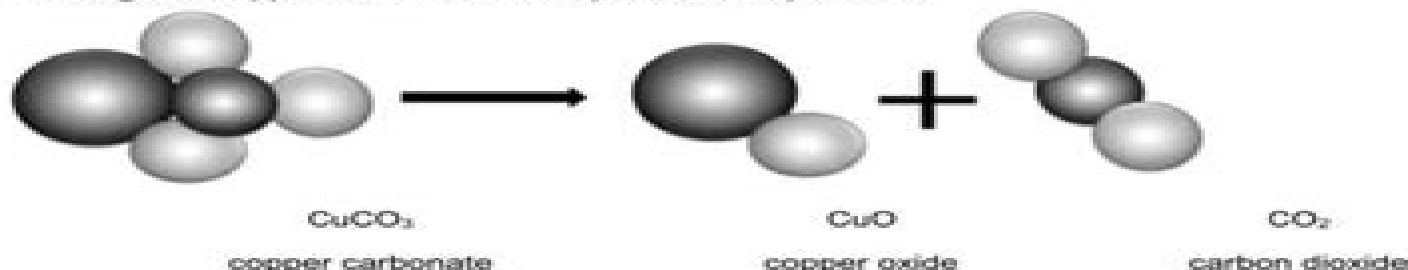
2 Making and breaking bonds

During chemical reactions the bonds between atoms break and new bonds form.

Energy must be absorbed to break a bond, so breaking bonds is endothermic.

Making new bonds is exothermic because energy is released.

- a) When green copper carbonate decomposes, the equation is:



Is the reaction exothermic or endothermic? Use ideas about bonds to explain why.

__exo because two bonds are broken from 4 bonds

- b) Draw diagrams to show what happens when hydrogen reacts with oxygen. Mark the bonds broken in blue and the new bonds formed in red. The equation is:



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Joacim Rocklöv



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