

1. Atomic Number- 6
Atomic Mass- 12.01
Atomic Symbol- C
Element Name- Carbon
2. a) 37
b) 41
c) 24
d) 112
e) 14
f) 8
g) 9
3. a) 266 amu
b) 231.04 amu
c) 39.95 amu
d) 83.70 amu
e) 131.29 amu
f) 222 amu
g) 207.2 amu
- 4.
- 5.
6. Ionization energy is the amount of energy required to remove an electron from an element.
7. Electronegativity is a measure of the relative ability an element's atoms to attract electrons in a chemical bond.
8. A group is a vertical column of elements in the periodic table that have similar chemical properties.
9. A period is a horizontal row of elements in the period table.
10. As you move left to right on the periodic table, the ionization energy increases. As you move from top to bottom, the ionization energy decreases.
11. As you move left to right on the periodic table, the electronegativity increases. As you move from top to bottom, the electronegativity decreases. The noble gases are not included in this trend.

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Committee on Commerce, Science,
and Transportation**

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