

The Nucleus & Radioactive Decay – Ch. 22 (p.701-712)

PART A – MASS DEFECT & NUCLEAR BINDING ENERGY

1. Find the mass defect of oxygen-16 if the measured atomic mass is 15.994915 amu.
1 proton = 1.007276 amu, 1 neutron = 1.008665 amu, and 1 electron = 0.0005486 amu.

2. Calculate the nuclear binding energy of oxygen-16 given that 1 amu = 1.6605×10^{-27} kg.

PART B – NUCLEAR DECAY

Write equations for the following nuclear decay reactions. Make sure that both mass numbers and atomic numbers are balanced on each side.

- | | |
|--|---|
| 3. Decay of polonium-218 by alpha (α) emission. | 5. Decay of chlorine-32 by positron (β^+) emission. |
| 4. Decay of carbon-14 by beta (β^-) emission. | 6. Decay of promethium-142 by electron capture. |

PART C – HALF-LIFE

7. The half-life of phosphorous-30 is 2.5 min. If you start with 35 g of phosphorus-30, how many grams would remain after 20.0 min?

8. The half-life of polonium-210 is 138.4 days. How many milligrams of polonium-210 remain after 415.2 days if you start with 2.0 mg of the isotope?

9. 20.0 g of a radioactive isotope are present at 1:00 p.m., and 5.0 g remain at 2:00 p.m.
 - a. How many half-lives have gone by? _____
 - b. How long is the half-life of the isotope? _____
 - c. Predict how many grams will be left at 2:30 p.m. _____

Radioactive Decay Worksheet D Answers

**Holt Rinehart & Winston, Holt,
Rinehart and Winston Staff**



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Nuclear Energy, 1985 *Cutnell & Johnson Physics* John D. Cutnell, David Young, Kenneth W. Johnson, Shane Stadler, 2022 The newly revised Twelfth Edition of Cutnell's Physics delivers an effective and accessible introduction to college and university physics. It contains easy-to-follow explanations of critical math and problem-solving concepts. From kinematics to work and energy, temperature, heat, electricity, magnetism, and optics, as well as foundational concepts in more advanced subjects like special relativity, Physics is the ideal introductory text for students from any background. The greatest strength of the text is the synergistic relationship it develops between problem-solving and conceptual understanding. The book lays emphasis on building relevance of physics in day-to-day living and highlights the physics principles that come into play. A wide range of applications that are biomedical in nature and others that deal with modern technology. **Holt**

Science and Technology 2002 Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2002 **Modern Science and the Book of Genesis** James William Skehan, National Science Teachers Association, 1986 Based on the premise that knowledge of evolutionary theory is essential for understanding the natural world, this document was designed to assist science teachers and others as they consider the issues that influence the teaching of evolution. The position is taken that there is no conflict between data and sound theories based on science and religious beliefs based on the Bible. Information and perspectives are presented under the topic headings of 1 The Genesis of Genesis 2 Early Science Interprets Genesis 3 New Data 4 Creationism versus Science and 5 Two Kinds of Knowledge. References are listed, and the National Science Teacher Association's position statement on the Inclusion of Nonscience Tenets in Science Instruction is included. ML

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